

ATOC1060

Our Changing Environment

Class 11

**Aerosol, Clouds
and Climate**

Jamison A. Smith

Laboratory for Atmospheric and Space Physics
University of Colorado



Overview

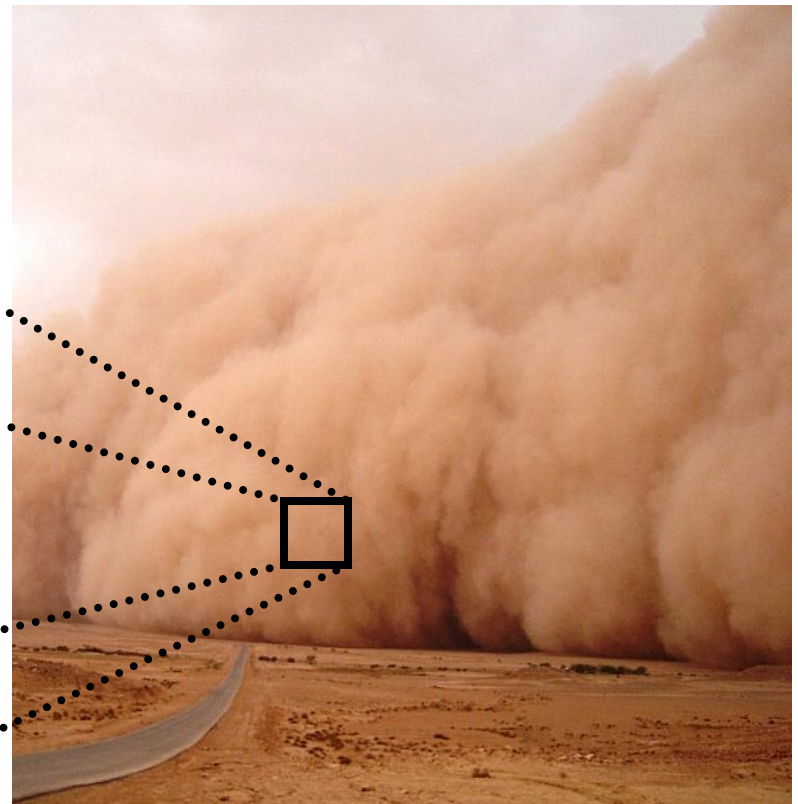
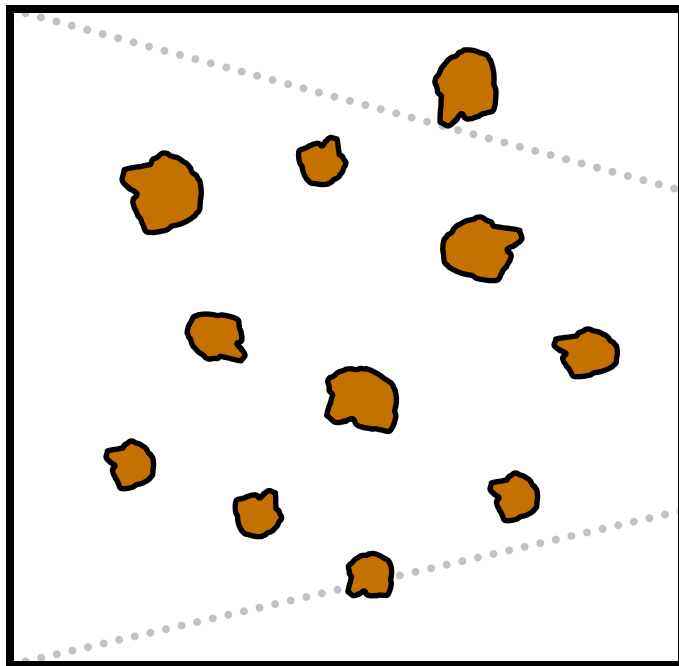
Objective: *to understand the impact of aerosol and clouds on climate*

- Introduce aerosol, types of aerosol and aerosol physics
- Introduce clouds and interactions among aerosol and clouds

Key terms: aerosol, scattering, direct aerosol effect, anthropogenic, absorption, semi-direct aerosol effect, indirect aerosol effects

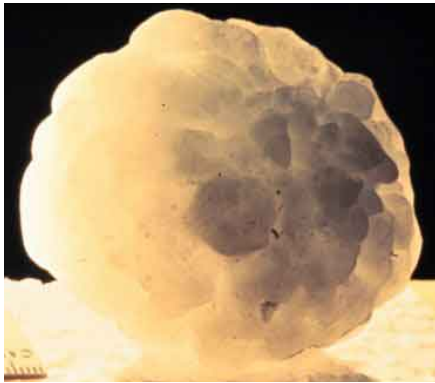
Aerosol

- An aerosol consists of particles suspended in a gas



Suspended Particles

Hailstone



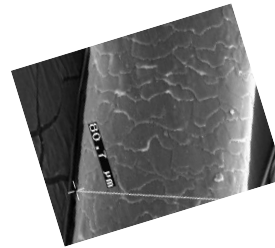
7" = 180 mm

Raindrop



5 mm

Human Hair



0.08 mm
= 80 μm

“micron”

Dust



2 μm

Sulfate Pollution



0.3 μm

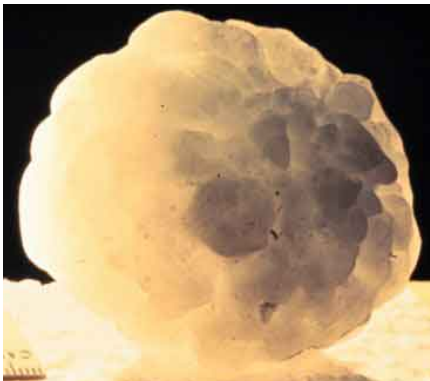
- Suspended temporarily by wind
- Eventually settle to the surface

Suspended Particles

Lifetime is also known as “residence time.”

It's the time required to transport the particle to the surface

Hailstone



minutes

Raindrop



minutes

Dust



**seconds
to days**

**Sulfate
Pollution**



**days to
years**

- Aerosol particles are difficult to simulate because of their short lifetimes; they are produced and removed rapidly

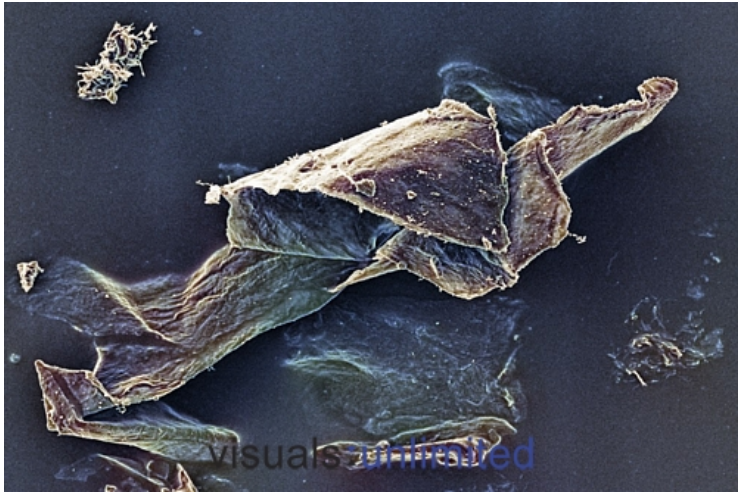
Suspended Particles

How do hailstones form?

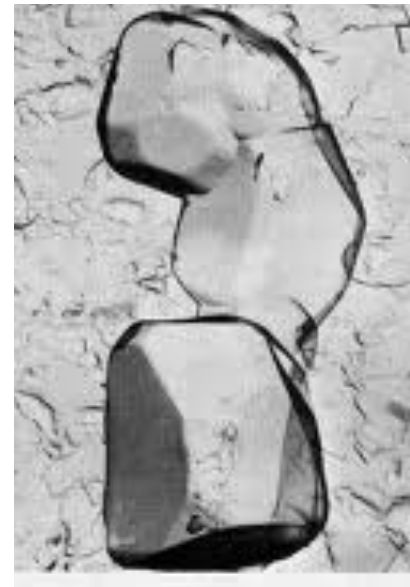
Demonstration

Aerosol Particles

What is dust?

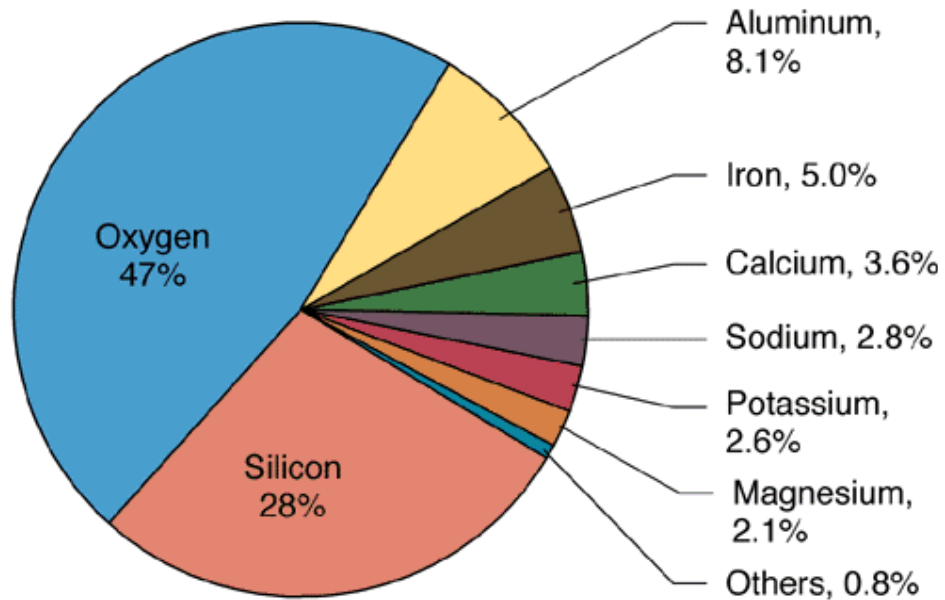


Household dust = shed skin



Wind-blown mineral dust

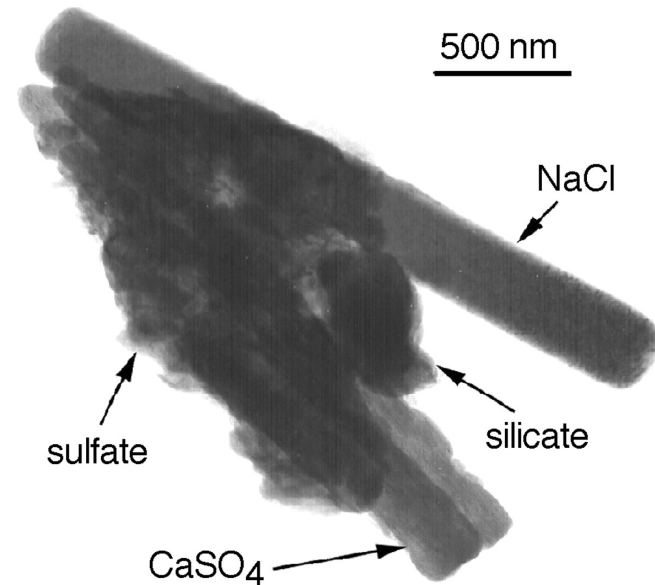
Dust Particles



SiO_2

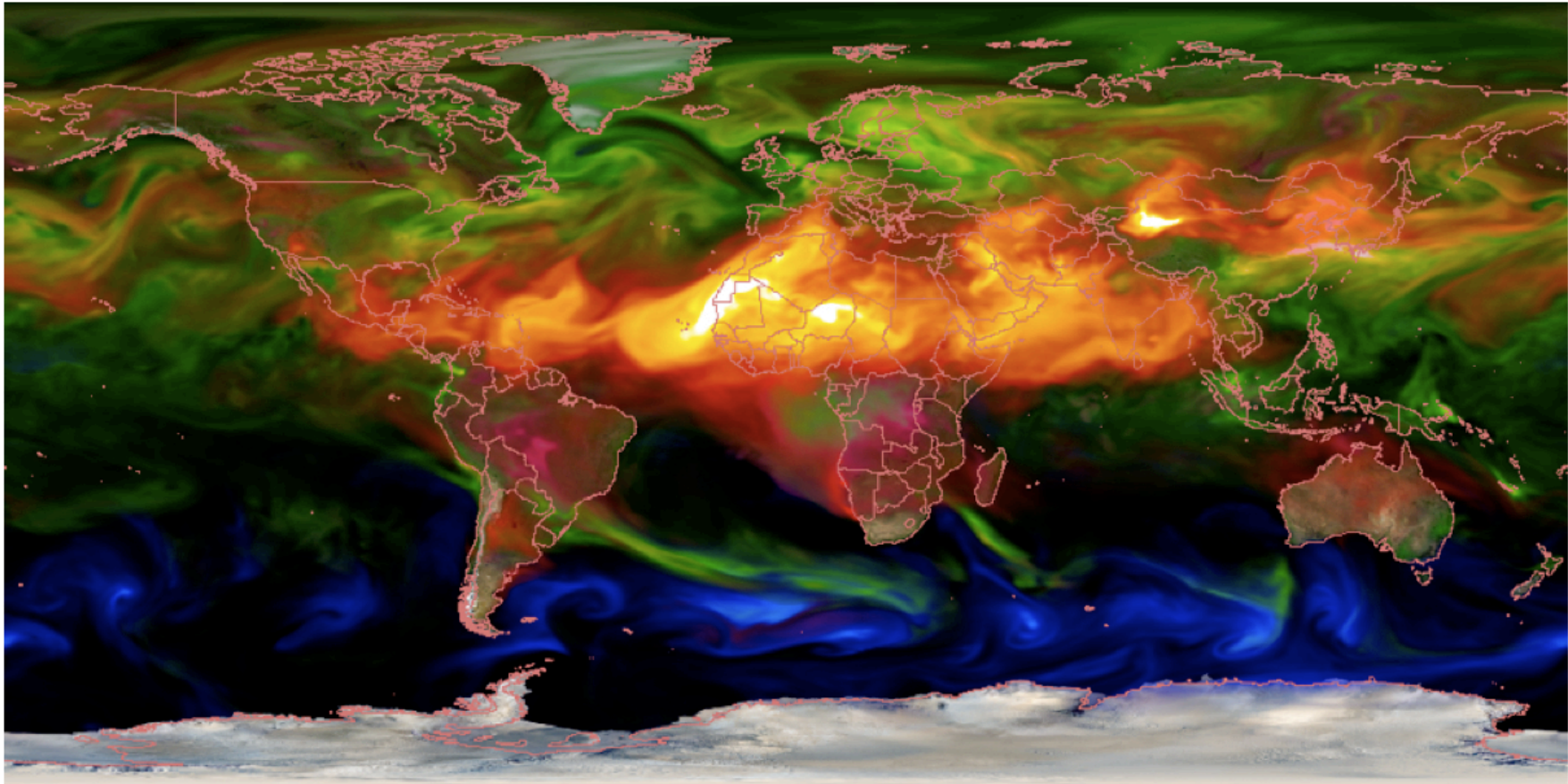


Al_2O_3



- Dust is mostly silicon dioxide, aluminum oxide and aluminosilicates, but there are a wide variety of minerals

Aerosol Types



Dust
Sulfate



Aerosol

Sea Salt
Carbonaceous

NASA GSFC Atmospheric Chemistry & Dynamics Branch, Global Modeling and Assimilation Office



“OK... I get aerosol already...”

Why are aerosol particles important for climate?



Beijing, China

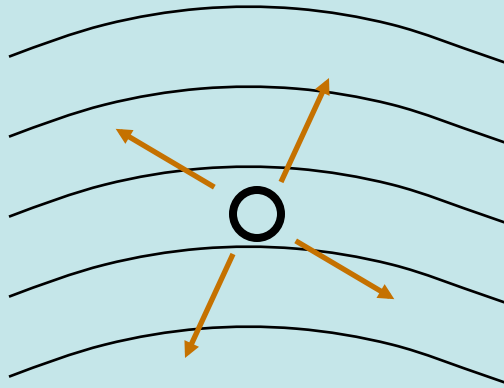


Beijing, China, 2002
Dust storm and pollution

- Aerosol particles affect the amount of sunlight that reaches the Earth -- breathability might be important, too

Are aerosol particles important for climate?

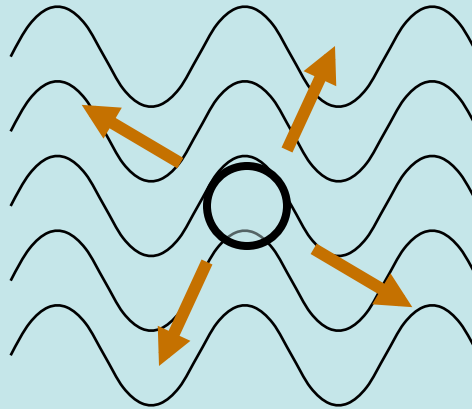
Aerosol Physics: Optics



Weakly scattering

Rayleigh Scattering

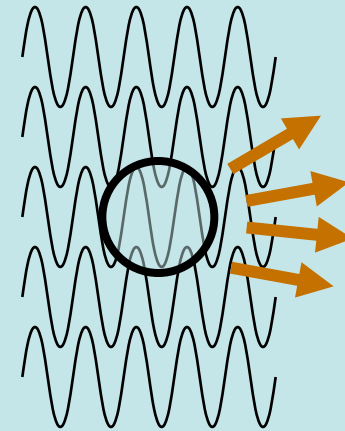
Clear Sky



Strongly scattering

Mie Scattering

Haze



Forward scattering

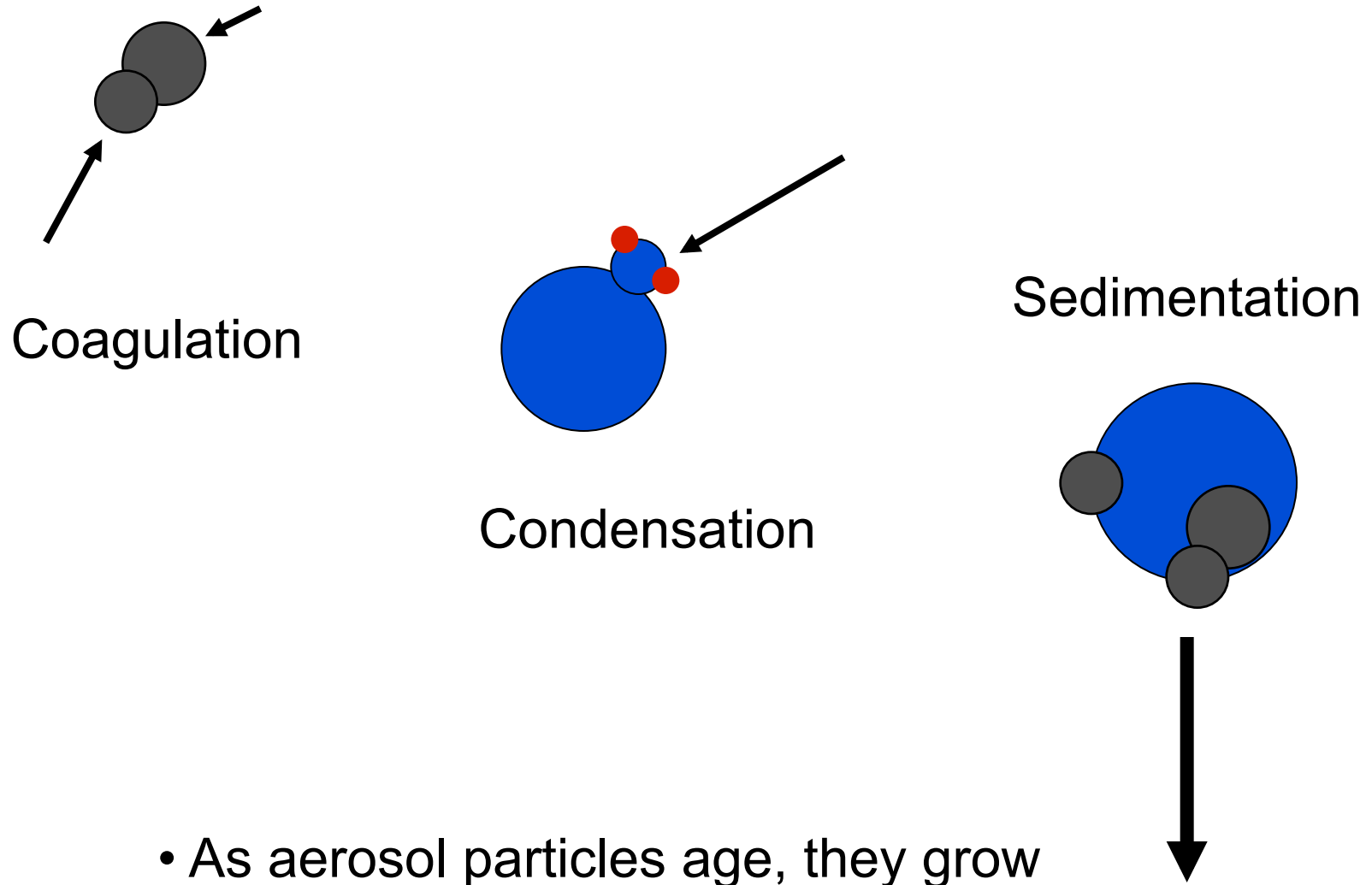
Geometric Optics

Dirty Windshield
(looking into the sun)

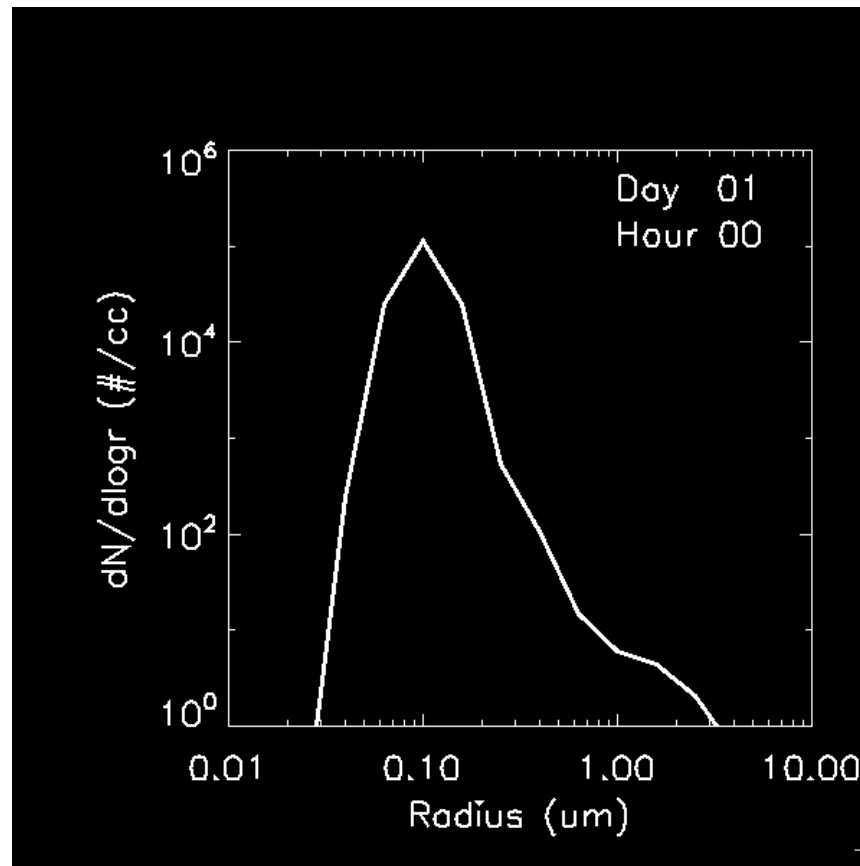
Scattering

- demonstration

Aerosol Microphysics

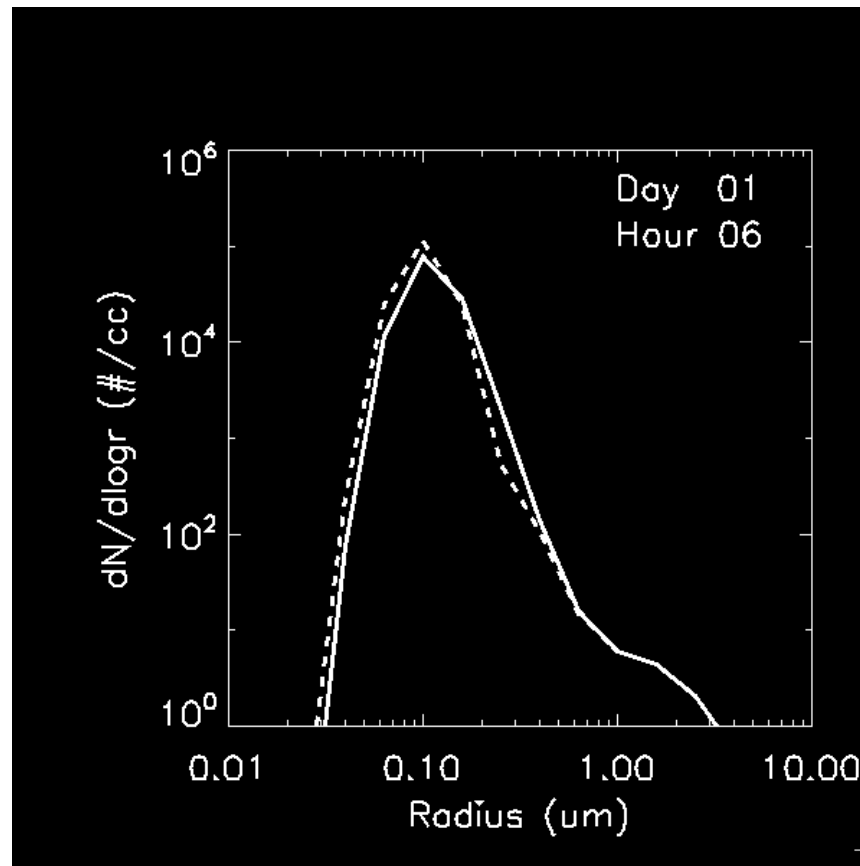


“Box of Smoke” Simulation



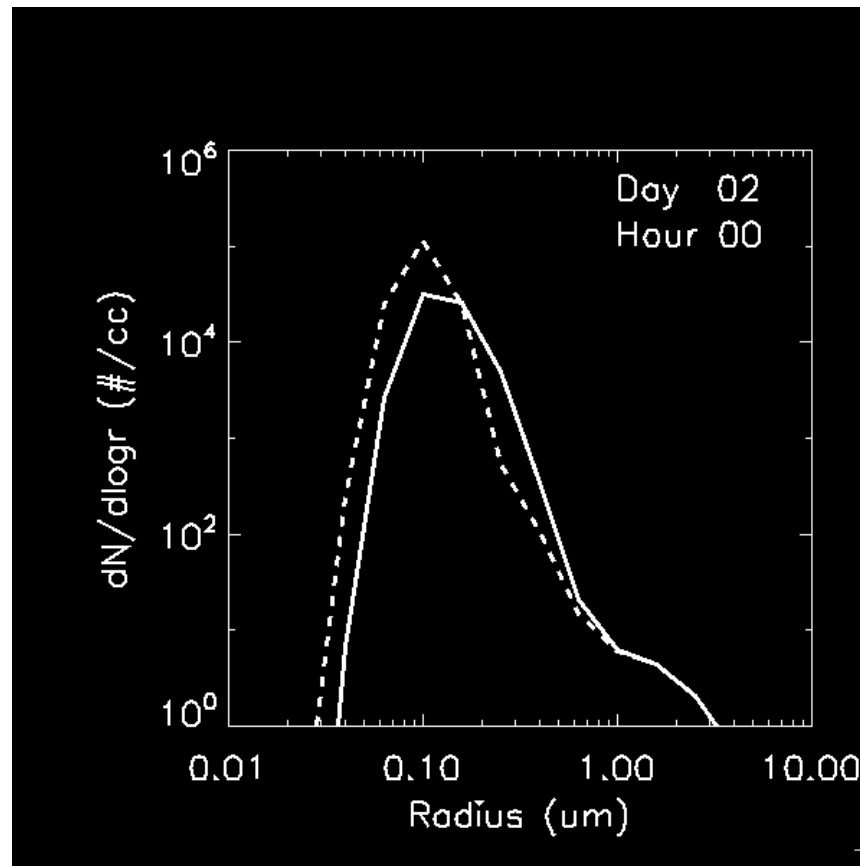
“Fresh” smoke from Haywood et al. JGR, 2003

“Box of Smoke” Simulation



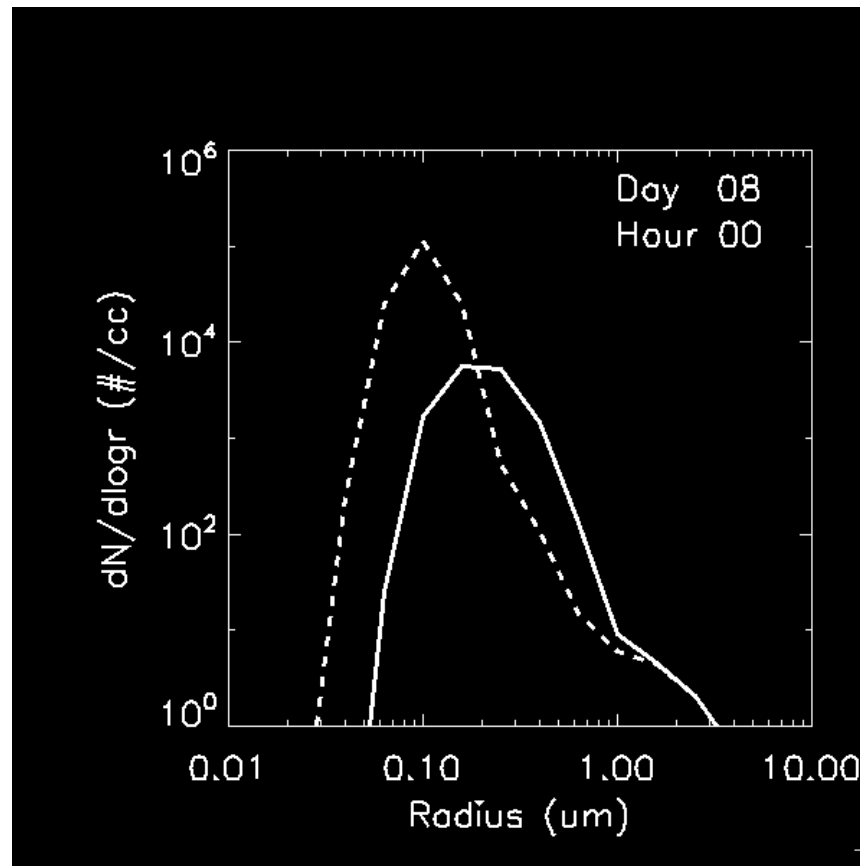
After 6 hours...

“Box of Smoke” Simulation



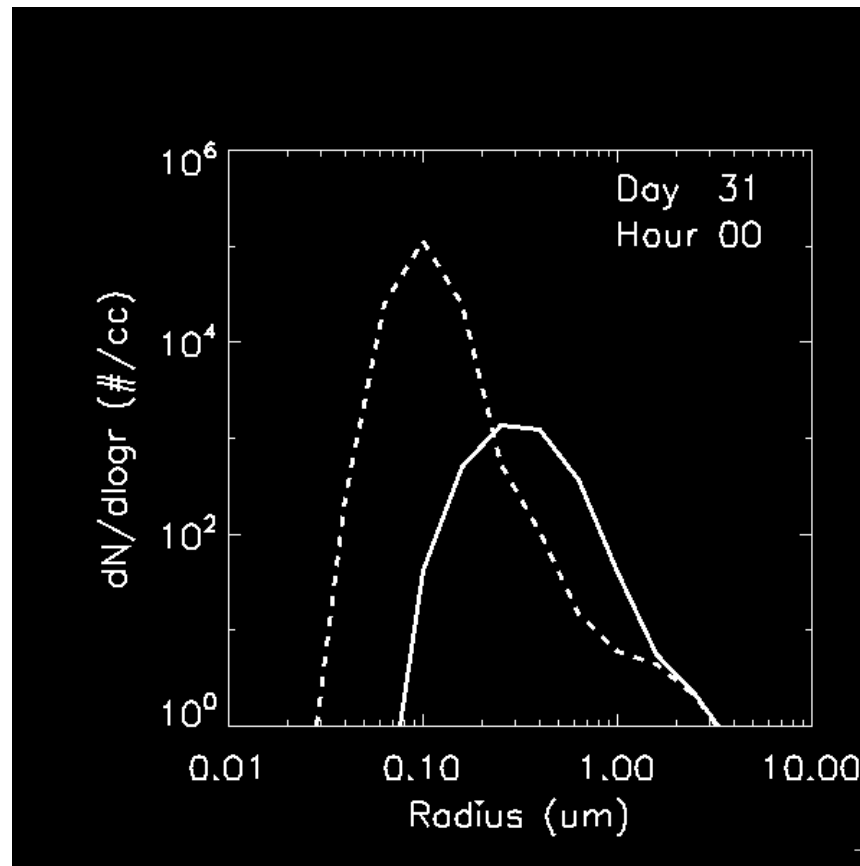
After one day...

“Box of Smoke” Simulation



One week old

“Box of Smoke” Simulation



- Remember. Scattering depends on particle size.

Coagulation and Scattering

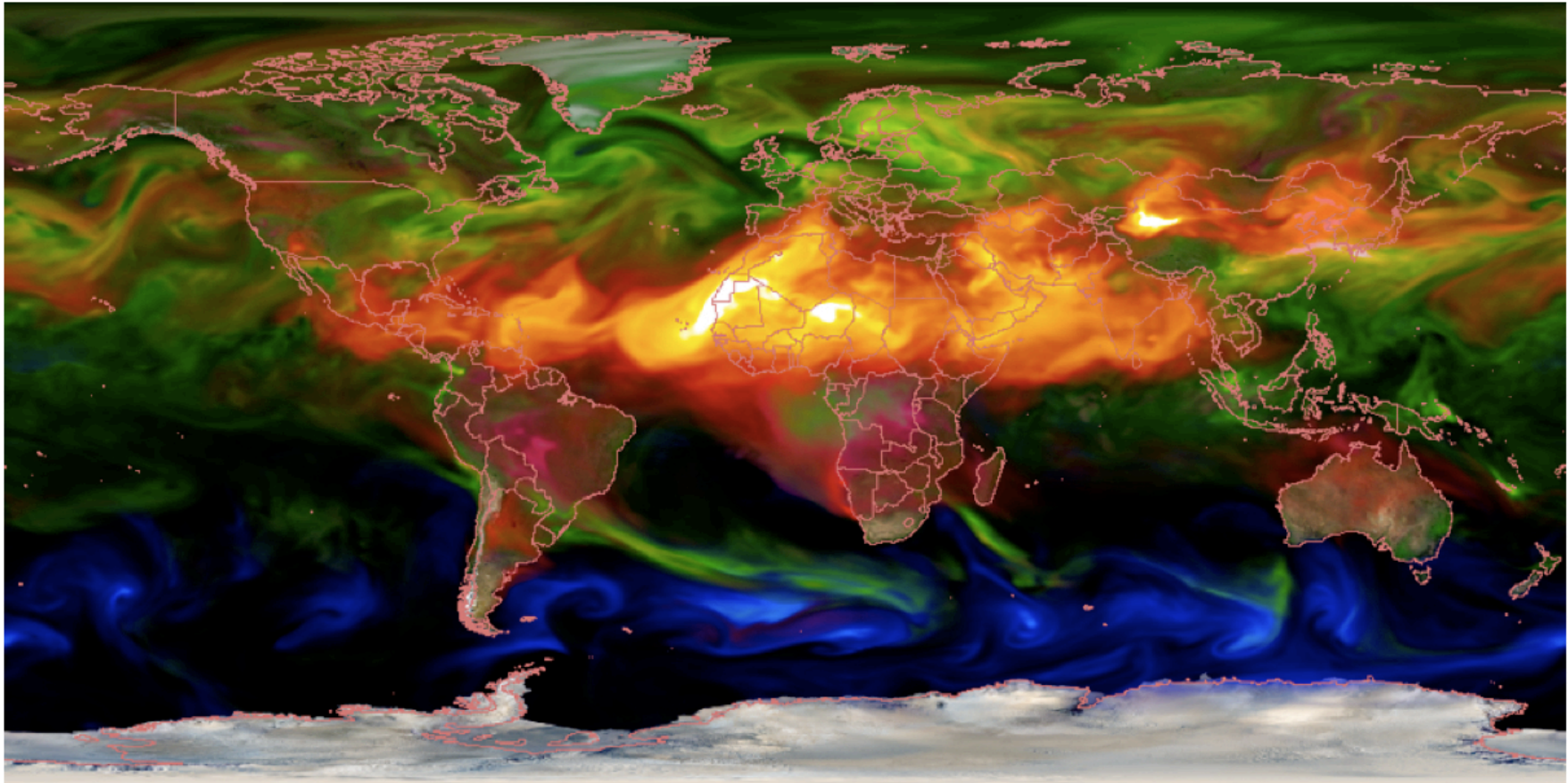


Left: Fresh smoke illuminated by white light. Particles appear “blue” because they are very small and narrowly distributed in size.



Right: After some time, the particles “coagulate” and grow in size. Particles appear white because they scatter all wavelengths about equally.

Aerosol and Climate



Dust
Sulfate



Aerosol

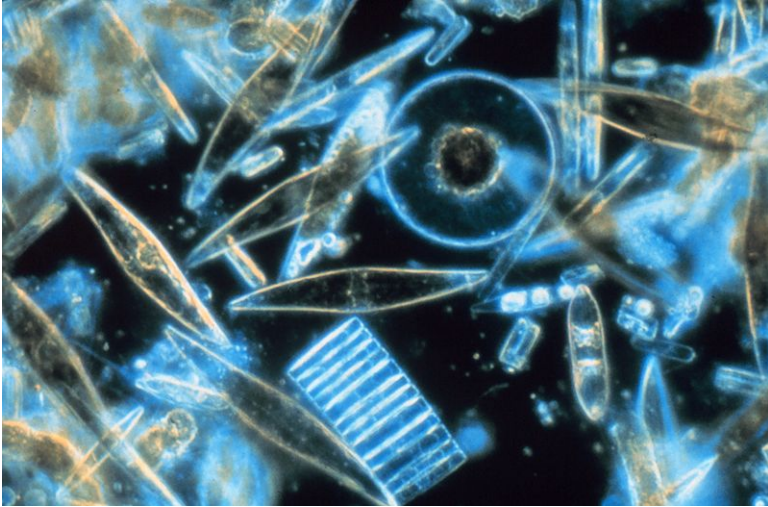
Sea Salt
Carbonaceous

NASA GSFC Atmospheric Chemistry & Dynamics Branch, Global Modeling and Assimilation Office



Sulfate Aerosol Sources

Natural (phytoplankton)

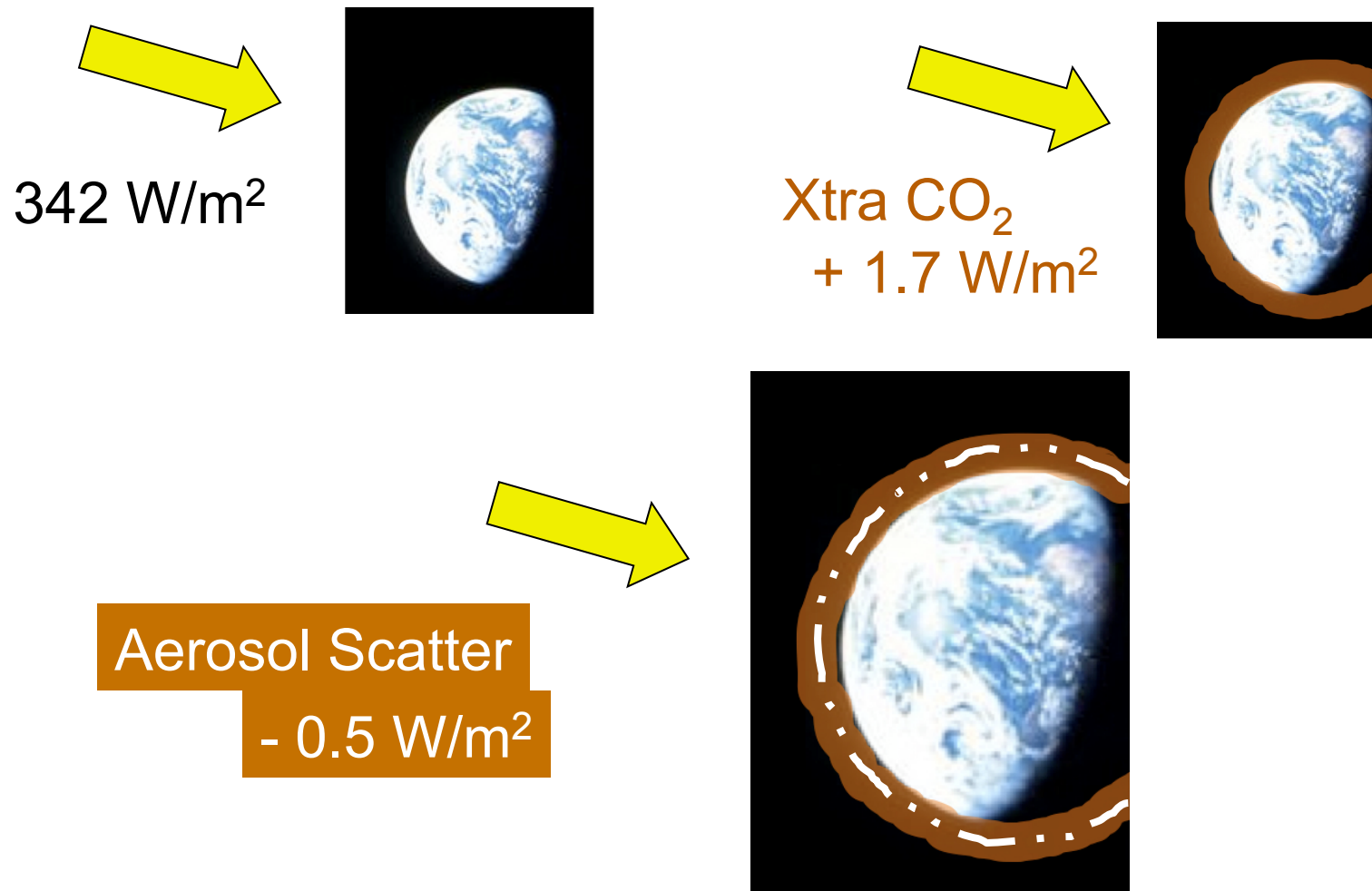


Anthropogenic (man-made)



- Both sources emit sulfur compounds
- These different sulfur compounds eventually oxidize to form sulfate aerosol ● $\sim 0.3 \mu\text{m}$
- Sulfate has many forms. Some are H_2SO_4 , NH_4HSO_4 and $(\text{NH}_4)_2\text{SO}_4$

The Direct Aerosol Effect



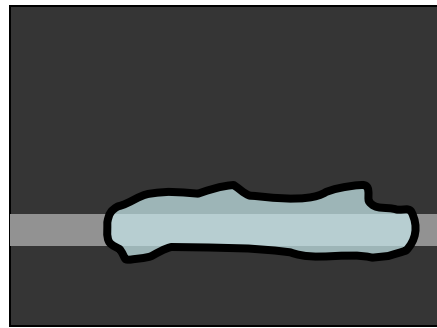
- Anthropogenic aerosol is counteracting greenhouse warming from anthropogenic CO₂

The Semi-Direct Aerosol Effect

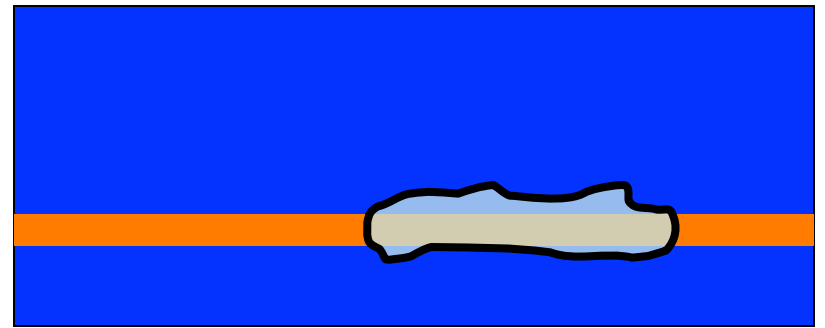
Smoke makes for an odd aerosol -- it both scatters and absorbs



Smoke layer at night does mostly nothing



Let's say a cloud forms during the night



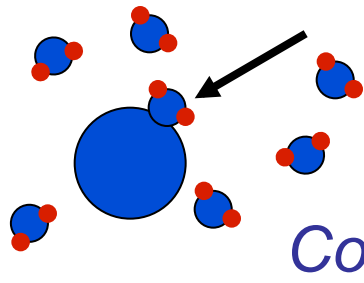
During the day, the smoke layer warms



Cloud evaporates prematurely

- Smoke evaporates existing clouds more quickly:
semi-direct aerosol effect

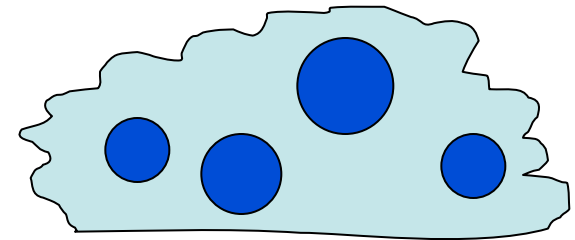
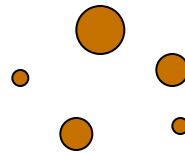
Clouds and the Indirect Aerosol Effect



Condensation

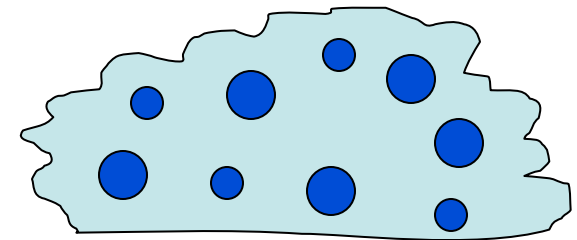
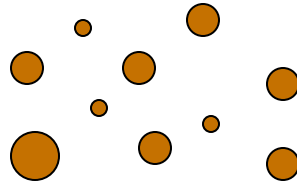
Clouds are made of aerosol haze particles that have “grown up”

Natural Aerosol
⇒ Cloud



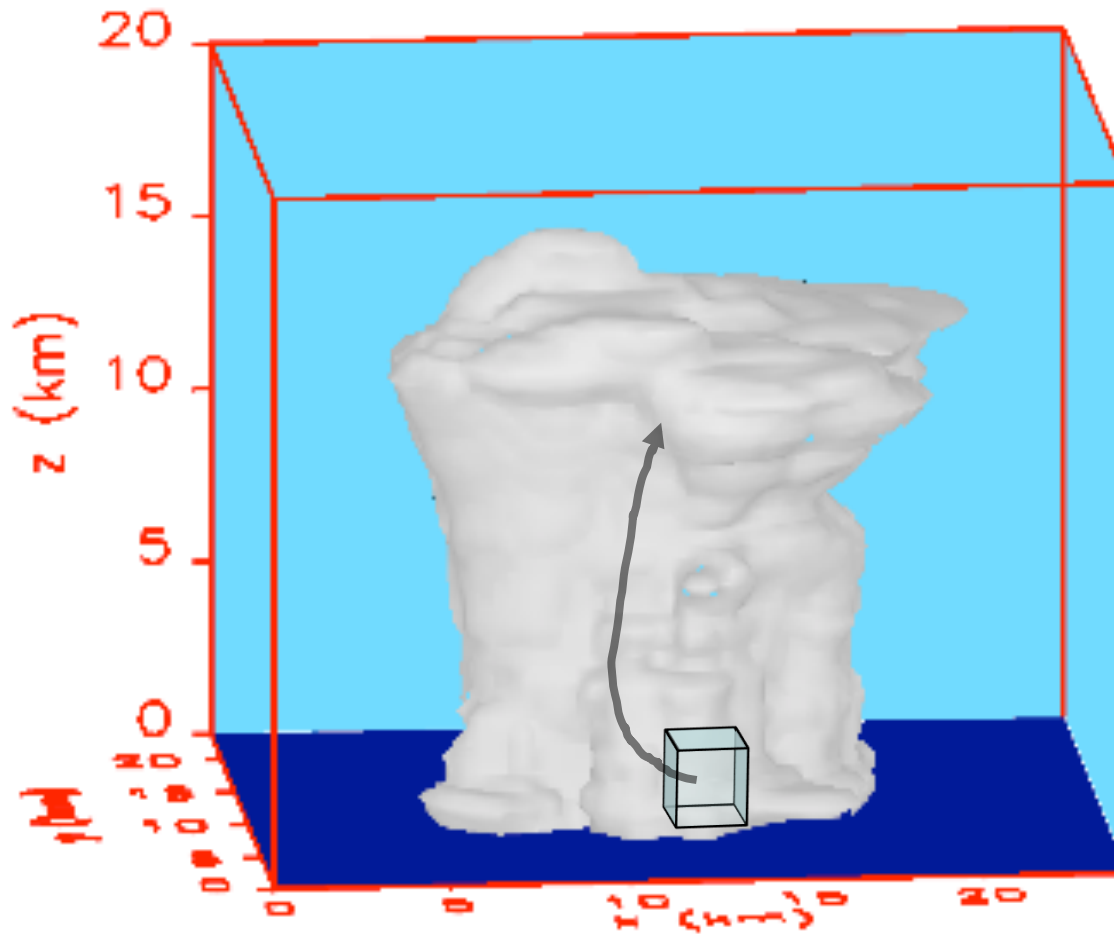
fewer, larger droplets

Polluted Aerosol
⇒ Cloud

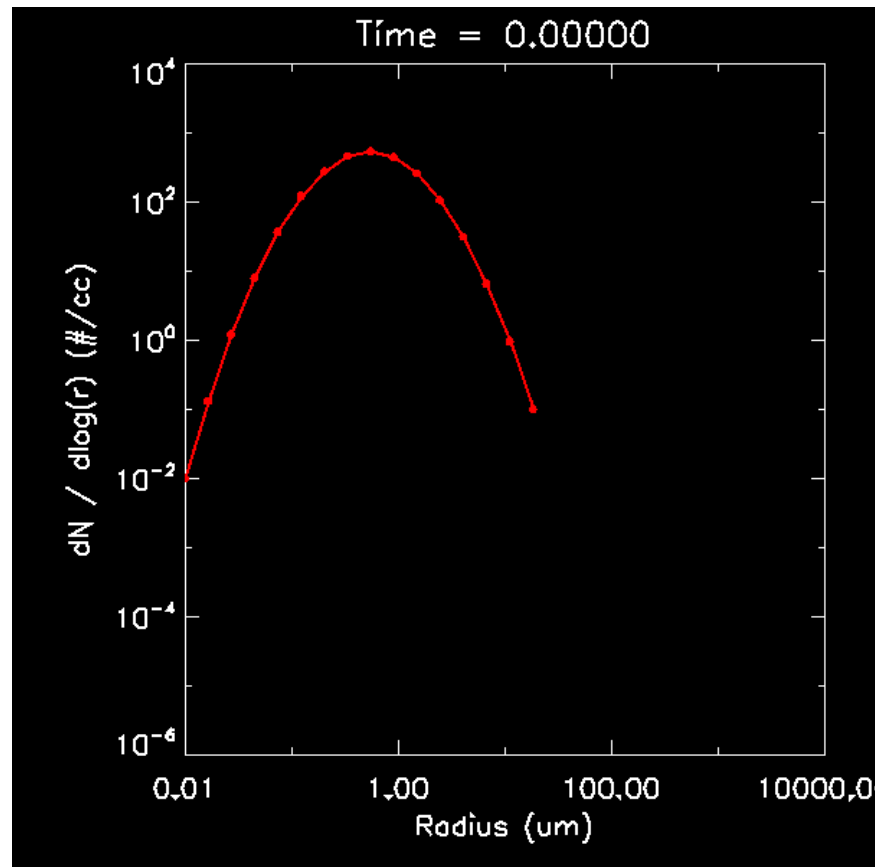


more, smaller droplets

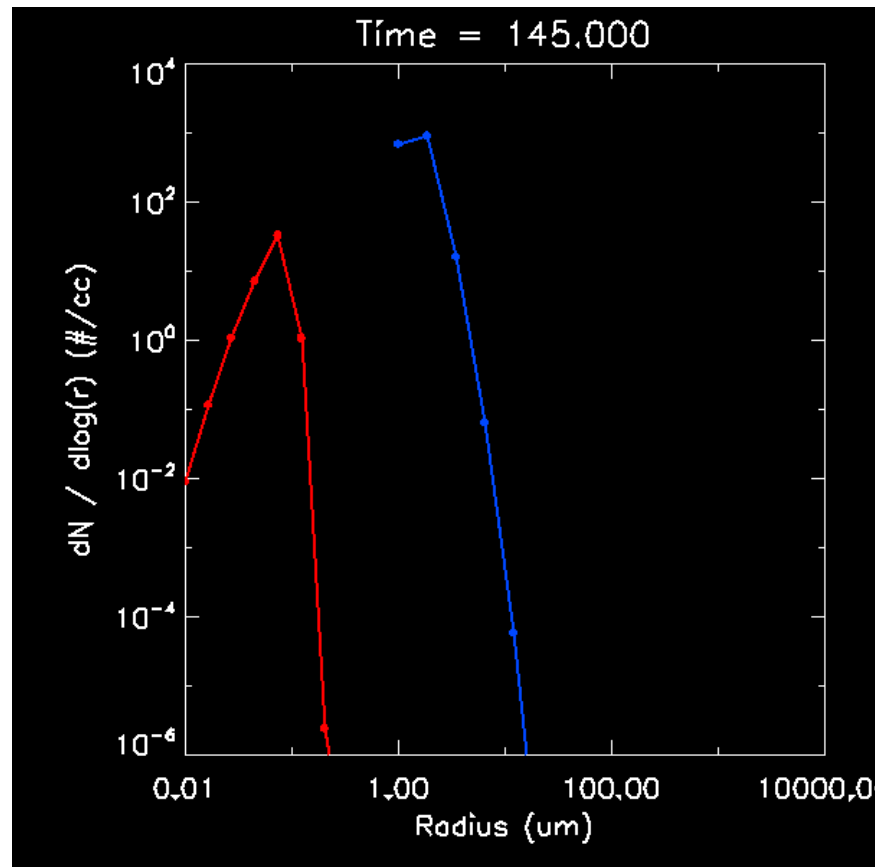
Deep Convection



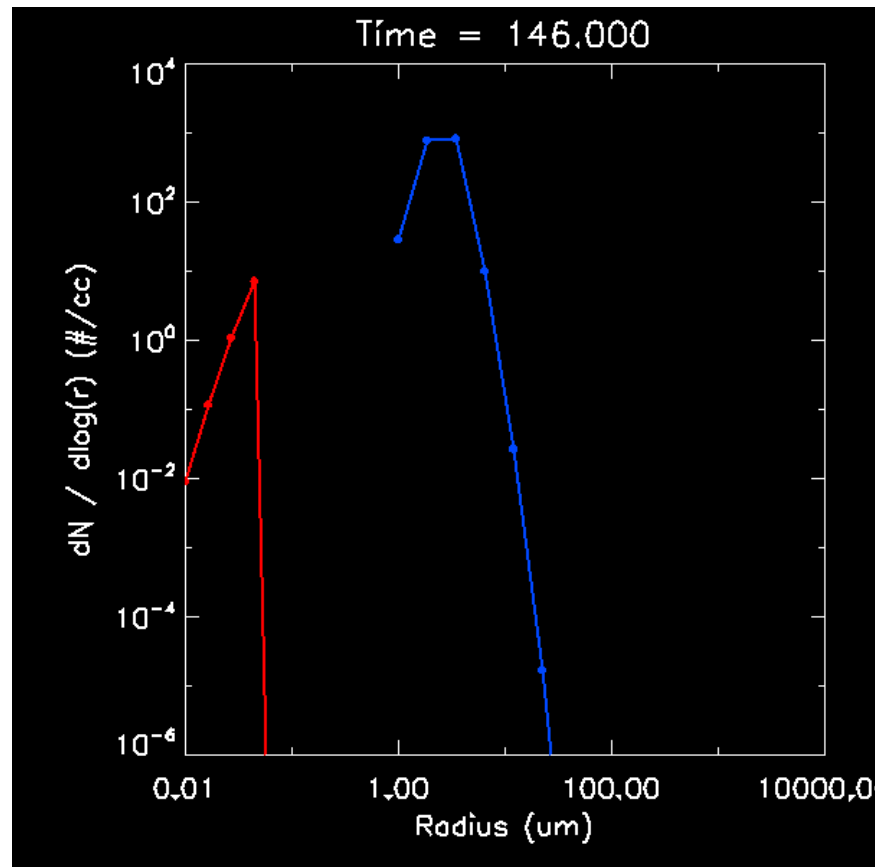
CARMA Parcel in an Updraft



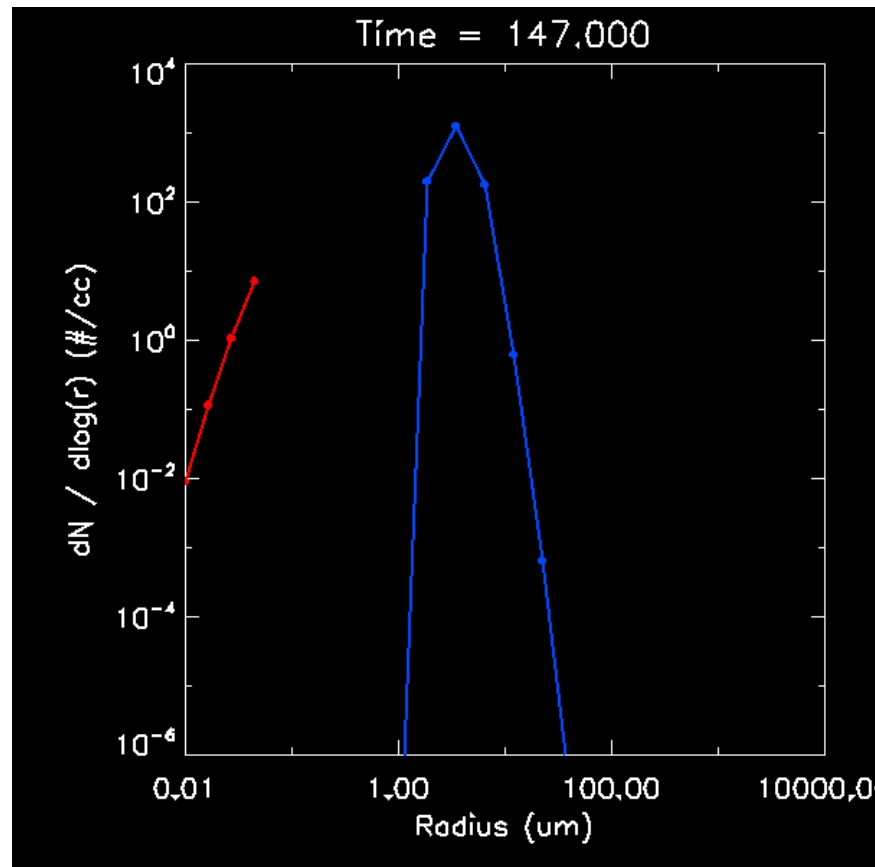
CARMA Parcel in an Updraft



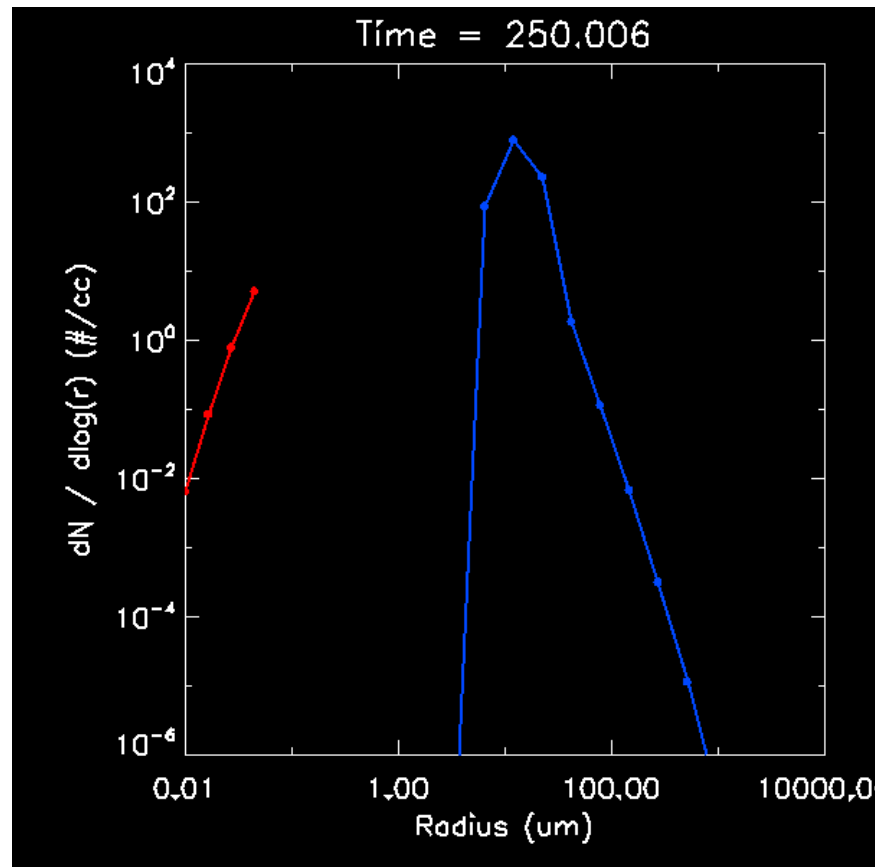
CARMA Parcel in an Updraft



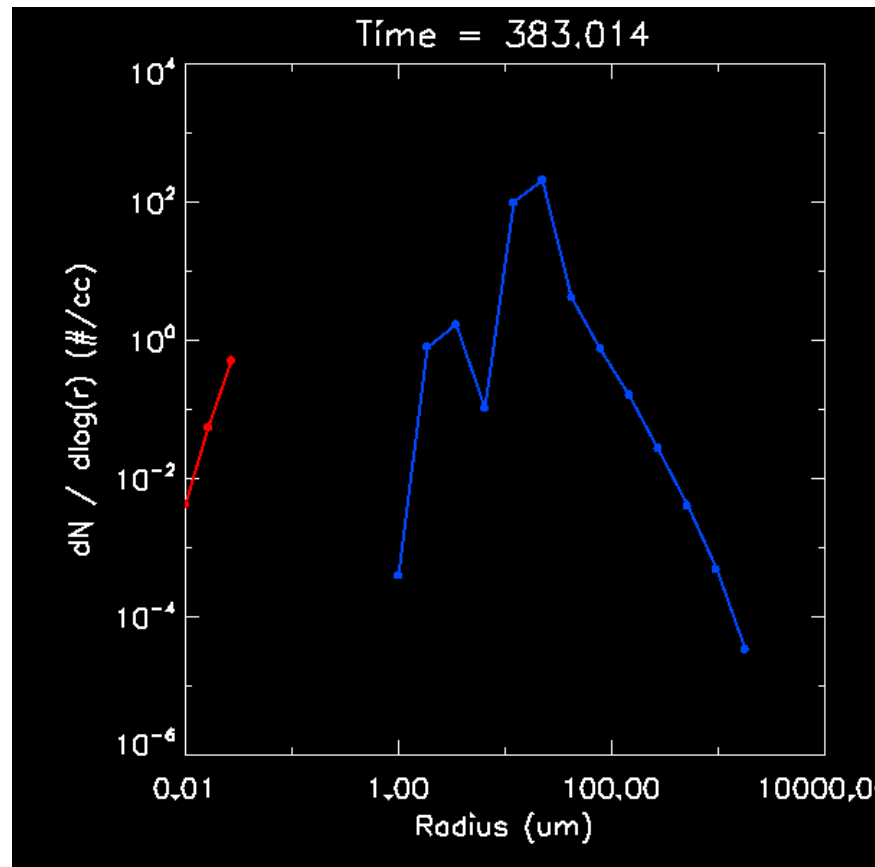
CARMA Parcel in an Updraft



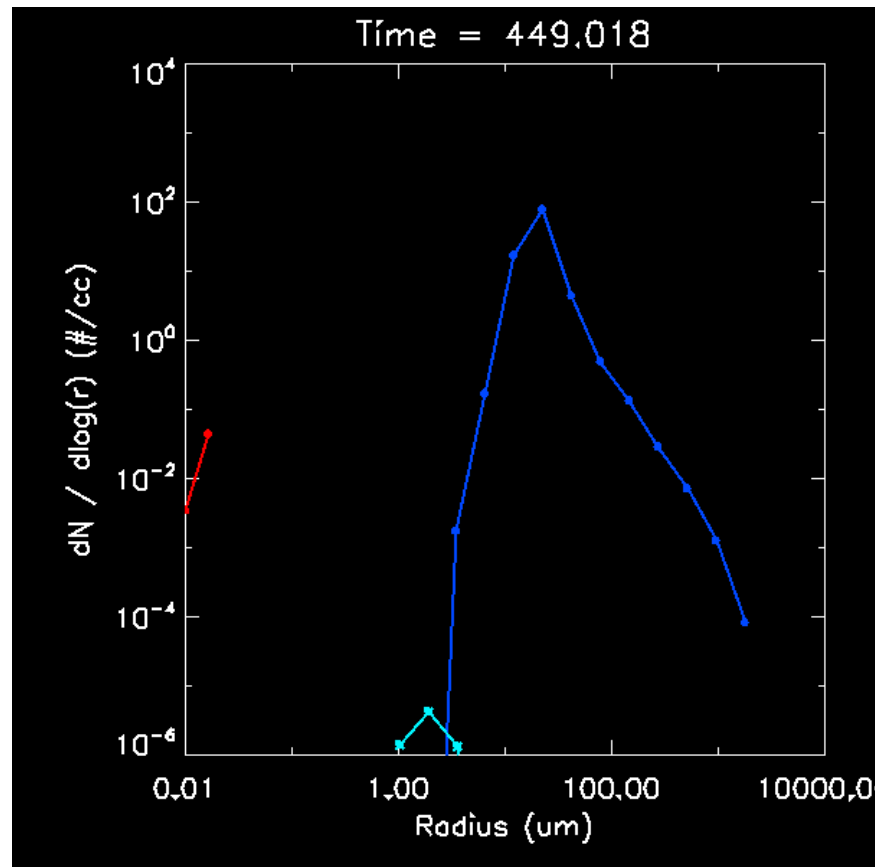
CARMA Parcel in an Updraft



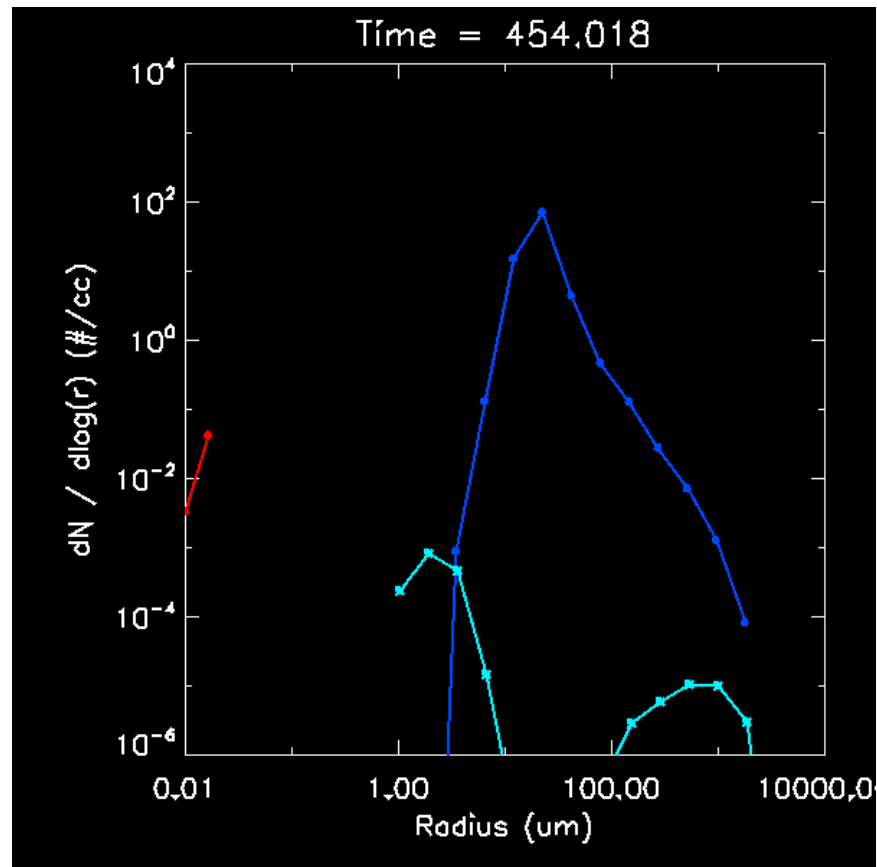
CARMA Parcel in an Updraft



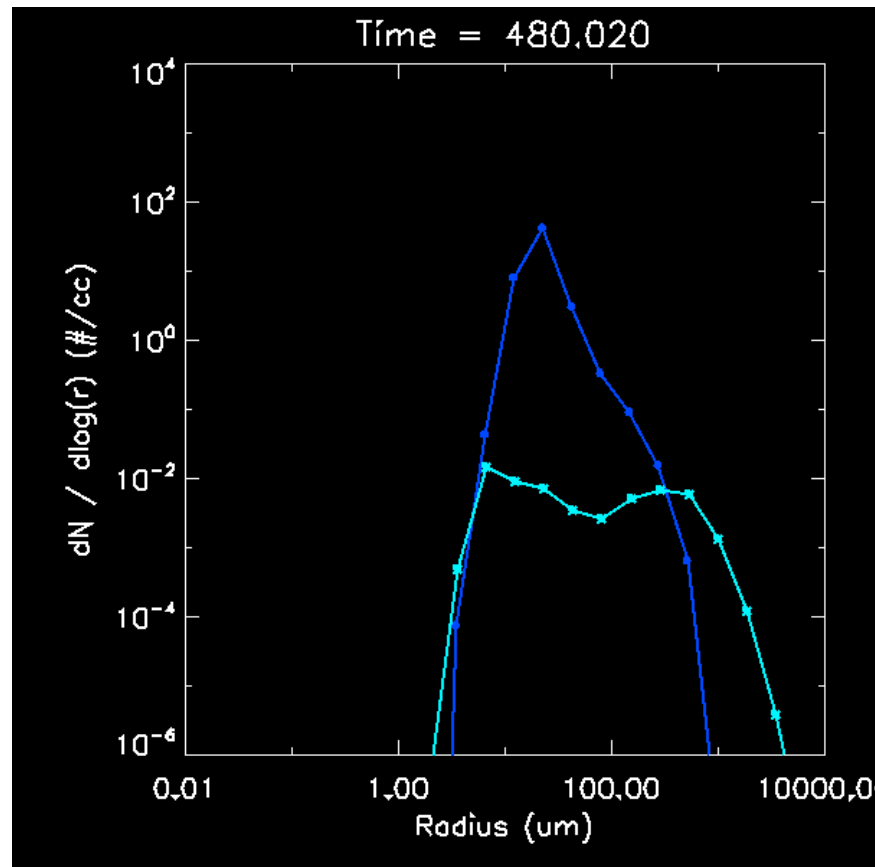
CARMA Parcel in an Updraft



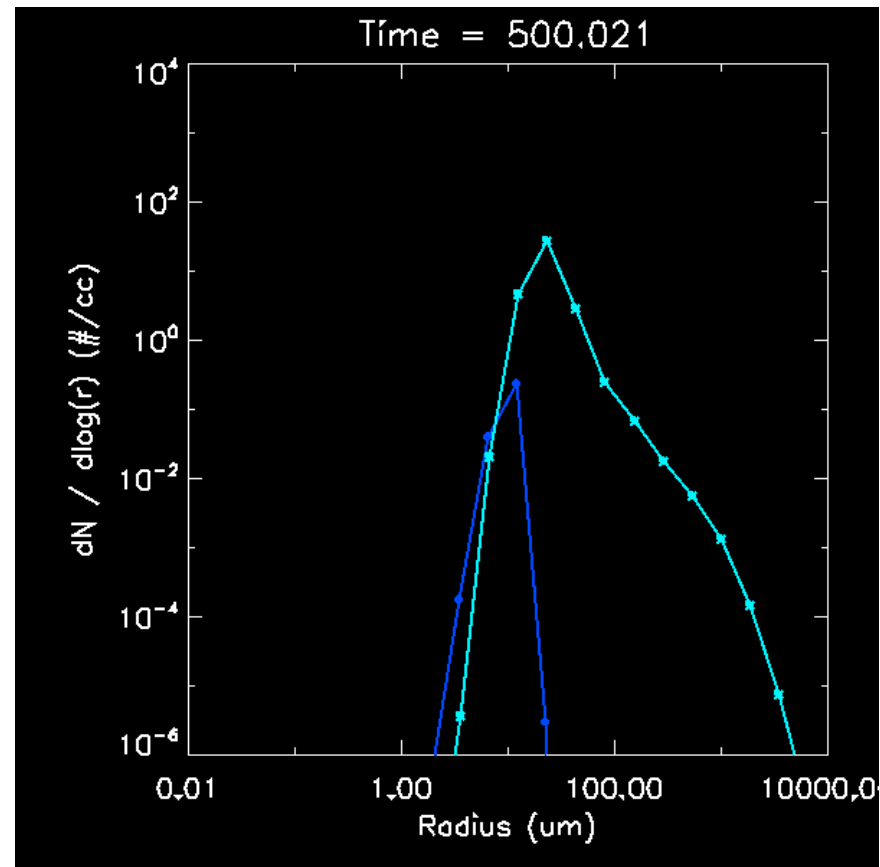
CARMA Parcel in an Updraft



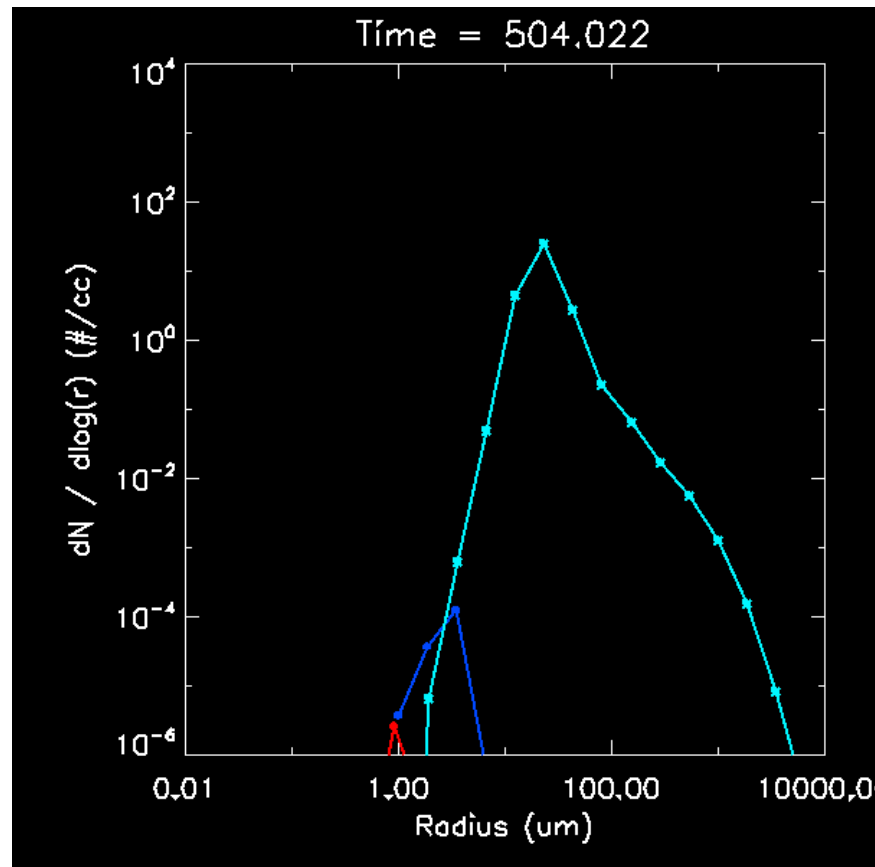
CARMA Parcel in an Updraft



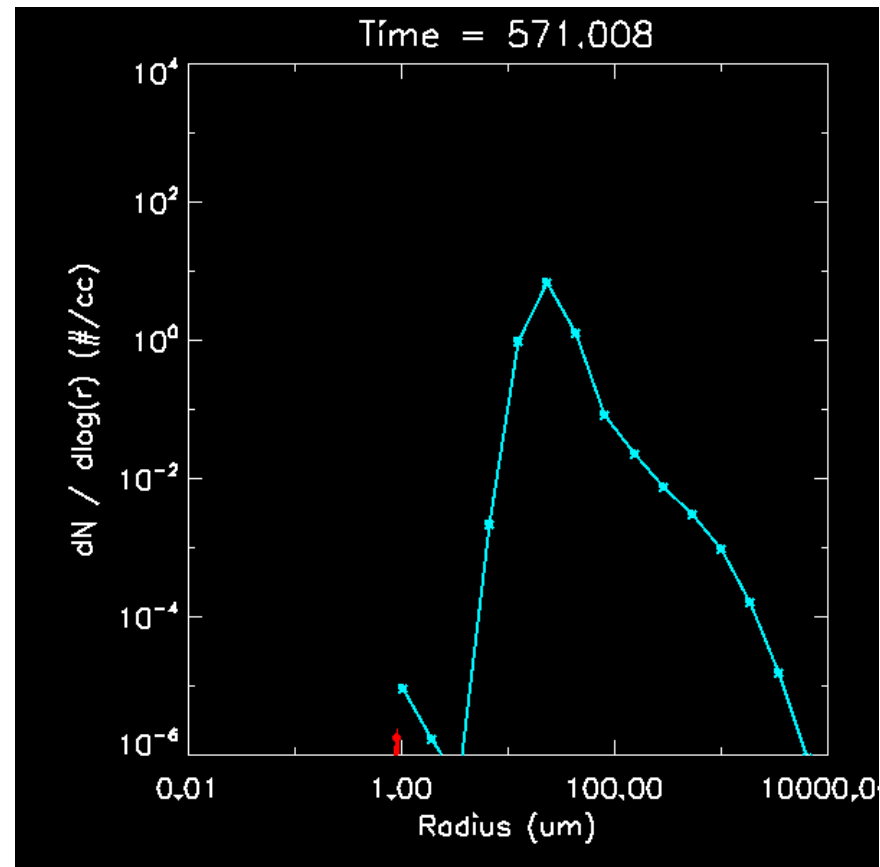
CARMA Parcel in an Updraft



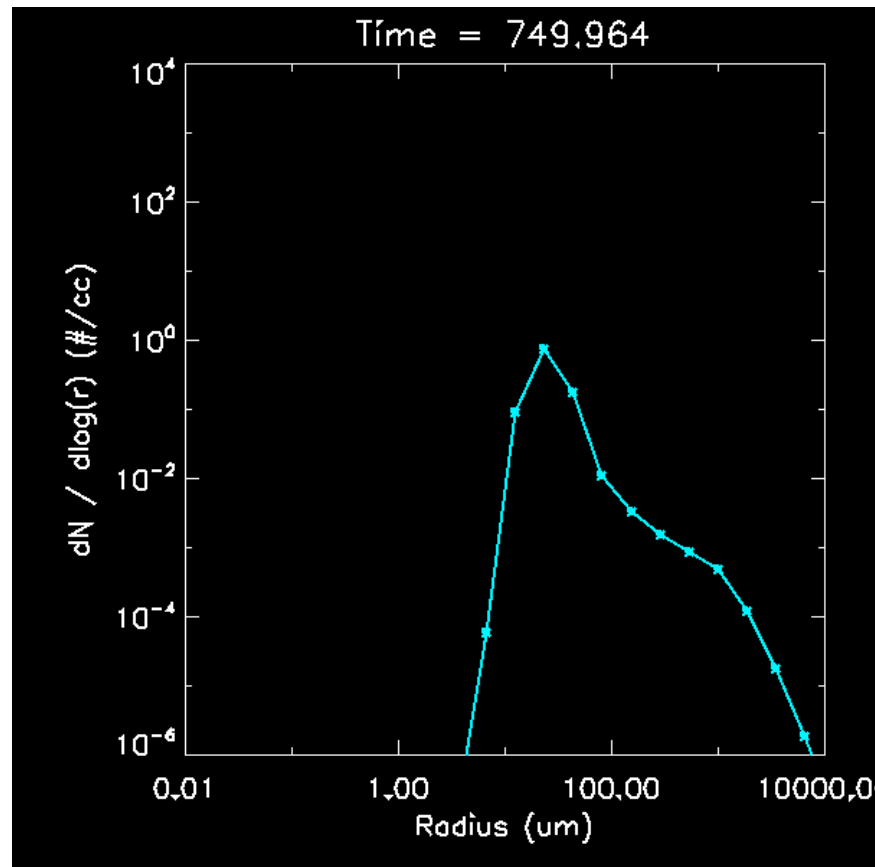
CARMA Parcel in an Updraft



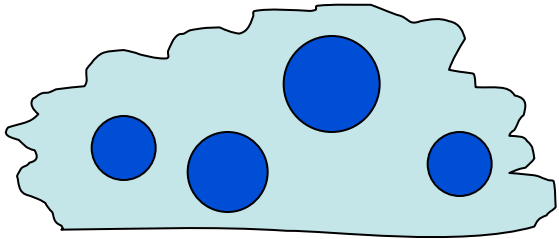
CARMA Parcel in an Updraft



CARMA Parcel in an Updraft

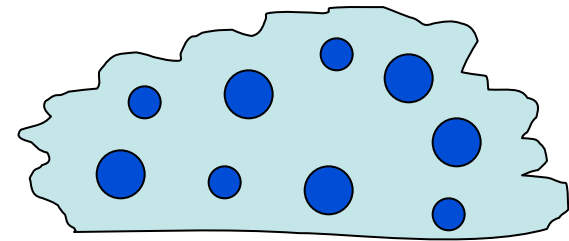
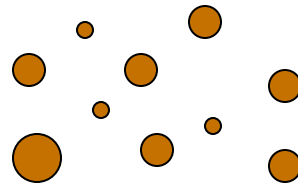


Clouds and the Indirect Aerosol Effect



fewer, larger droplets

Polluted Aerosol
⇒ Cloud

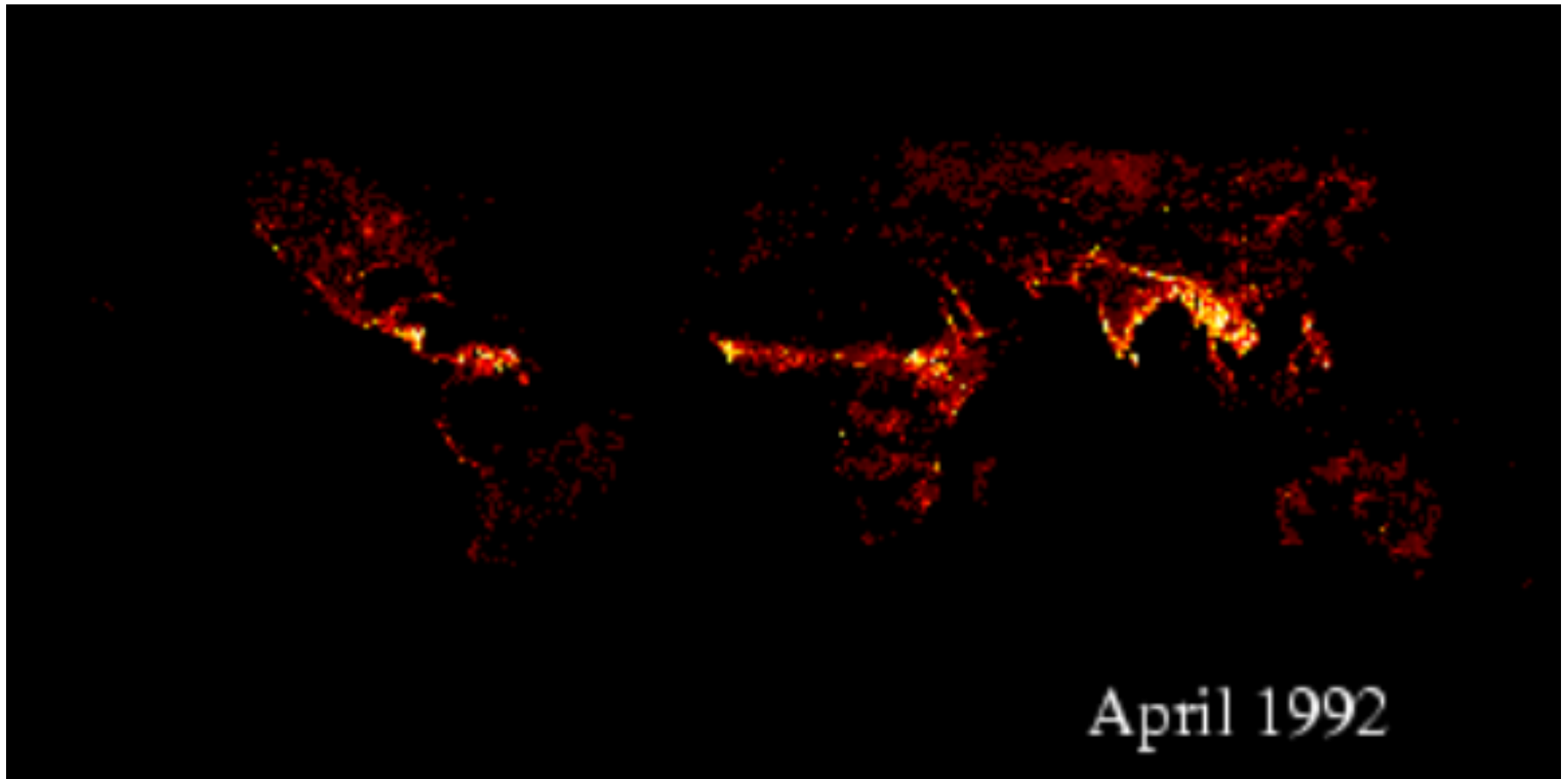


more, smaller droplets

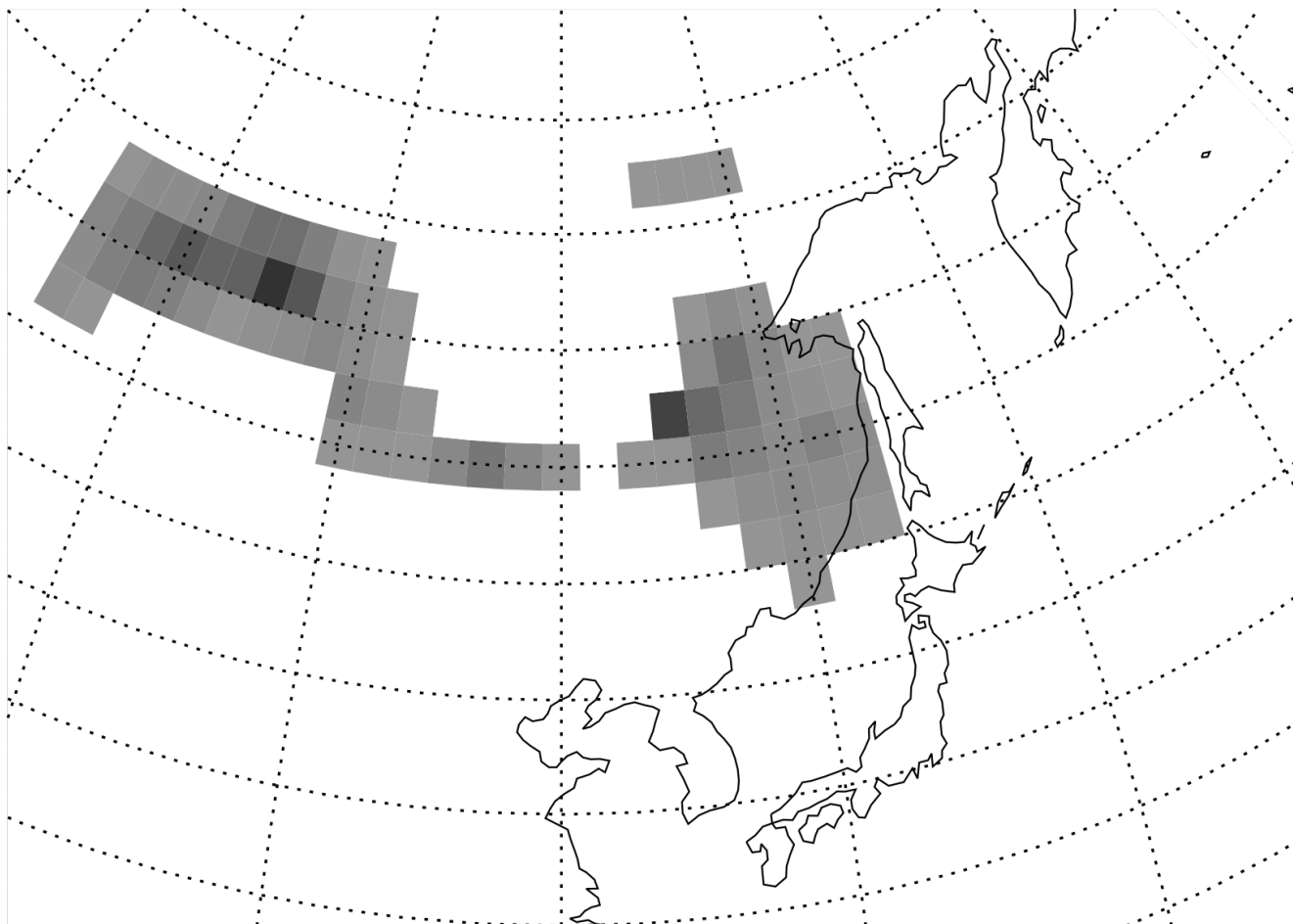
The Indirect Aerosol Effects

- Higher reflectivity
- Less drizzle
- Increased cloud height
- Increased cloud lifetime

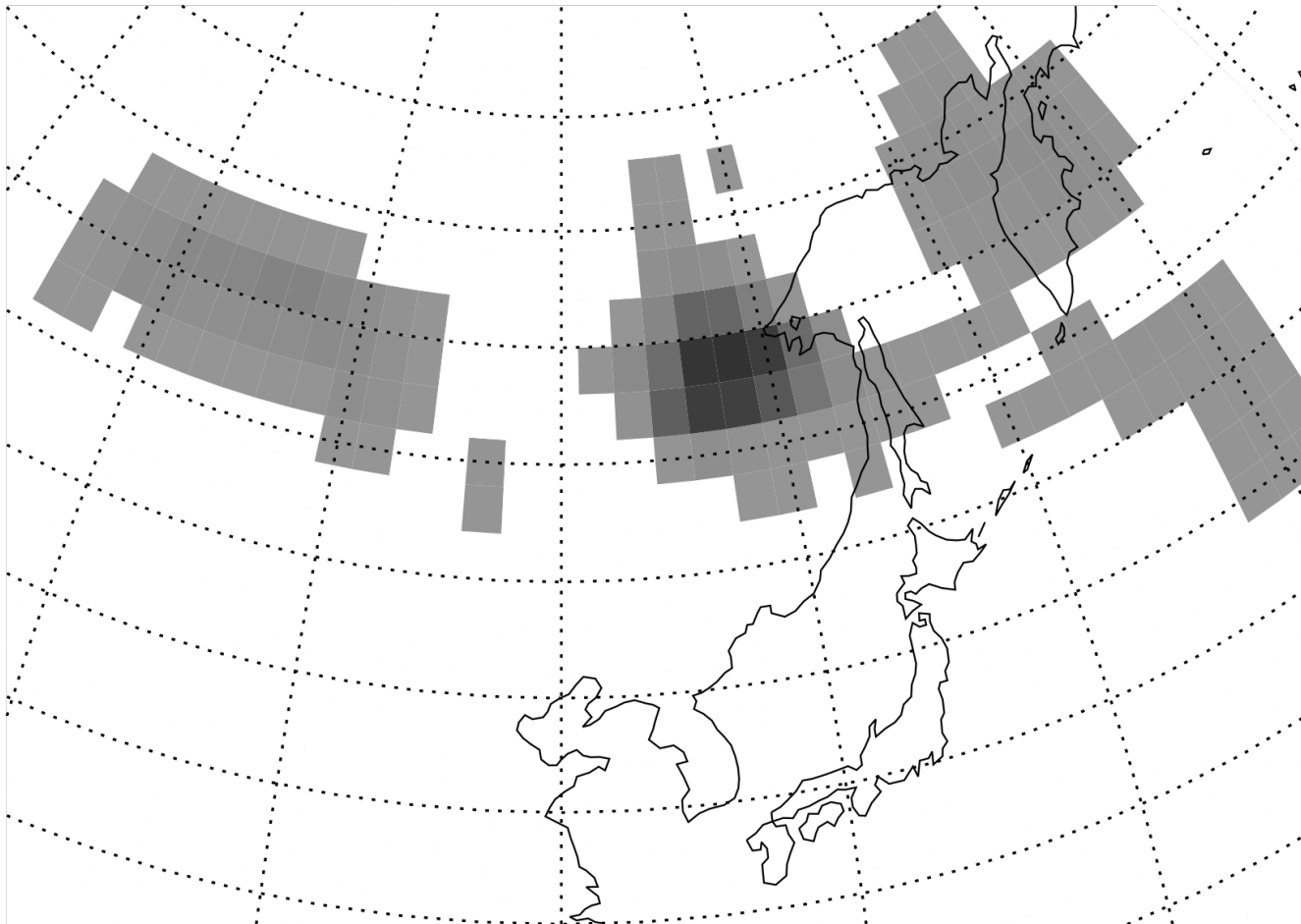
Fire and Deforestation



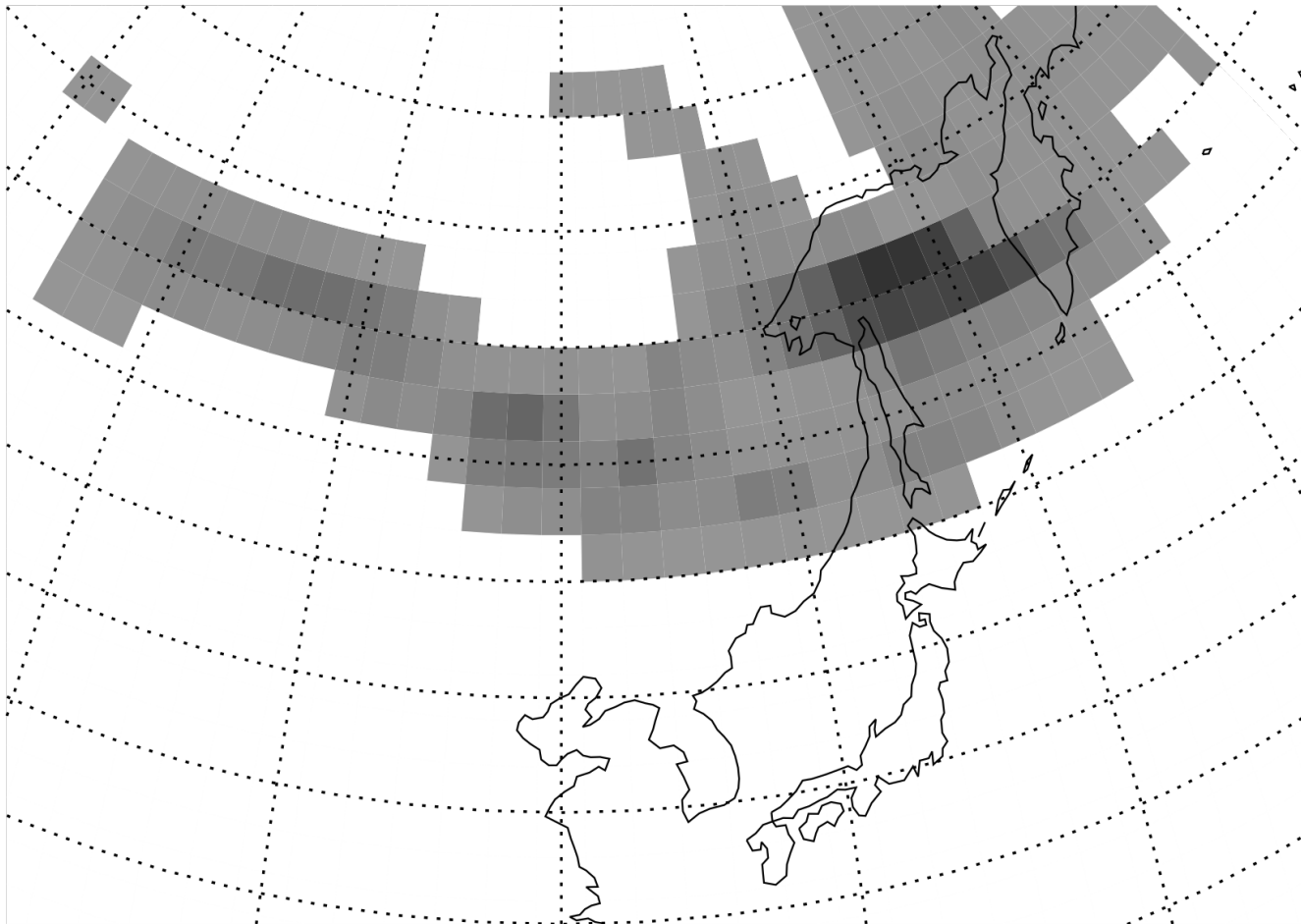
Siberian Smoke 20030501



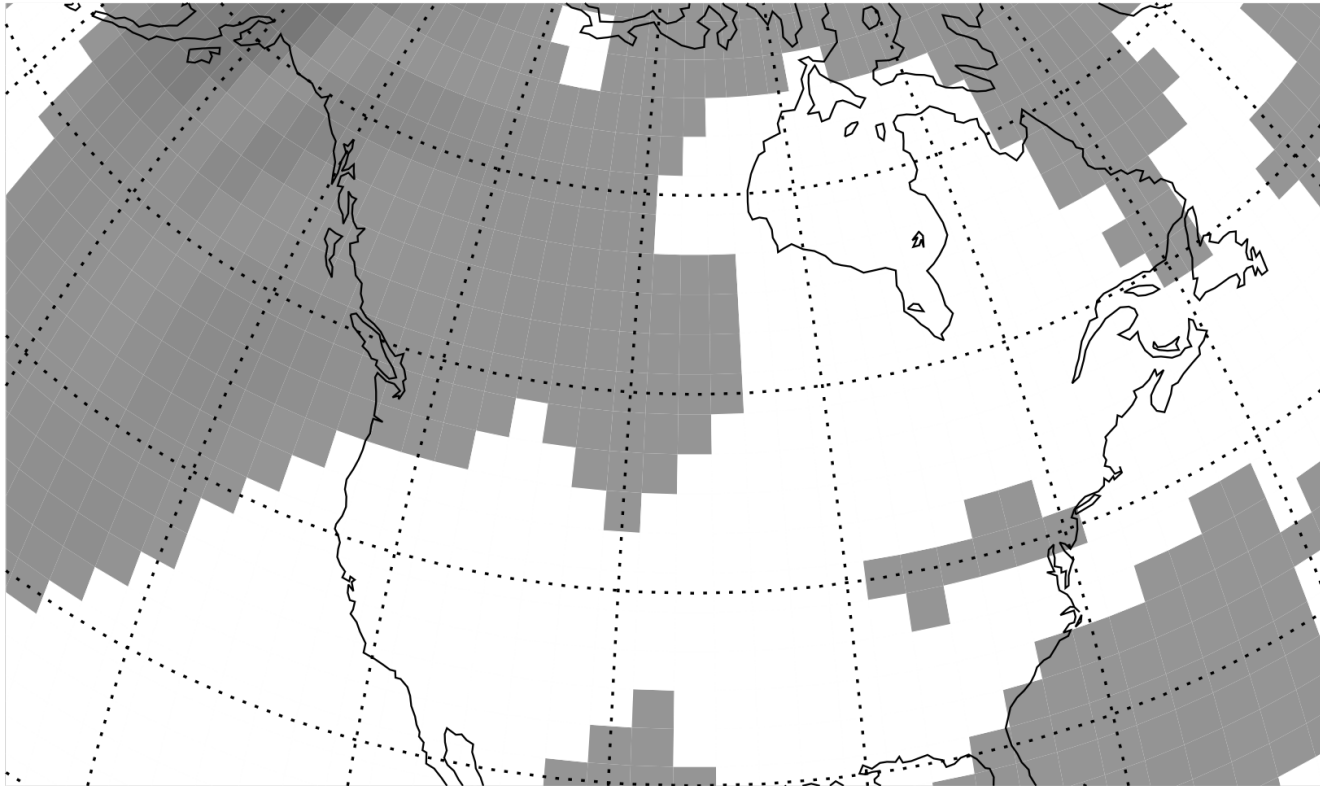
Siberian Smoke 20030502



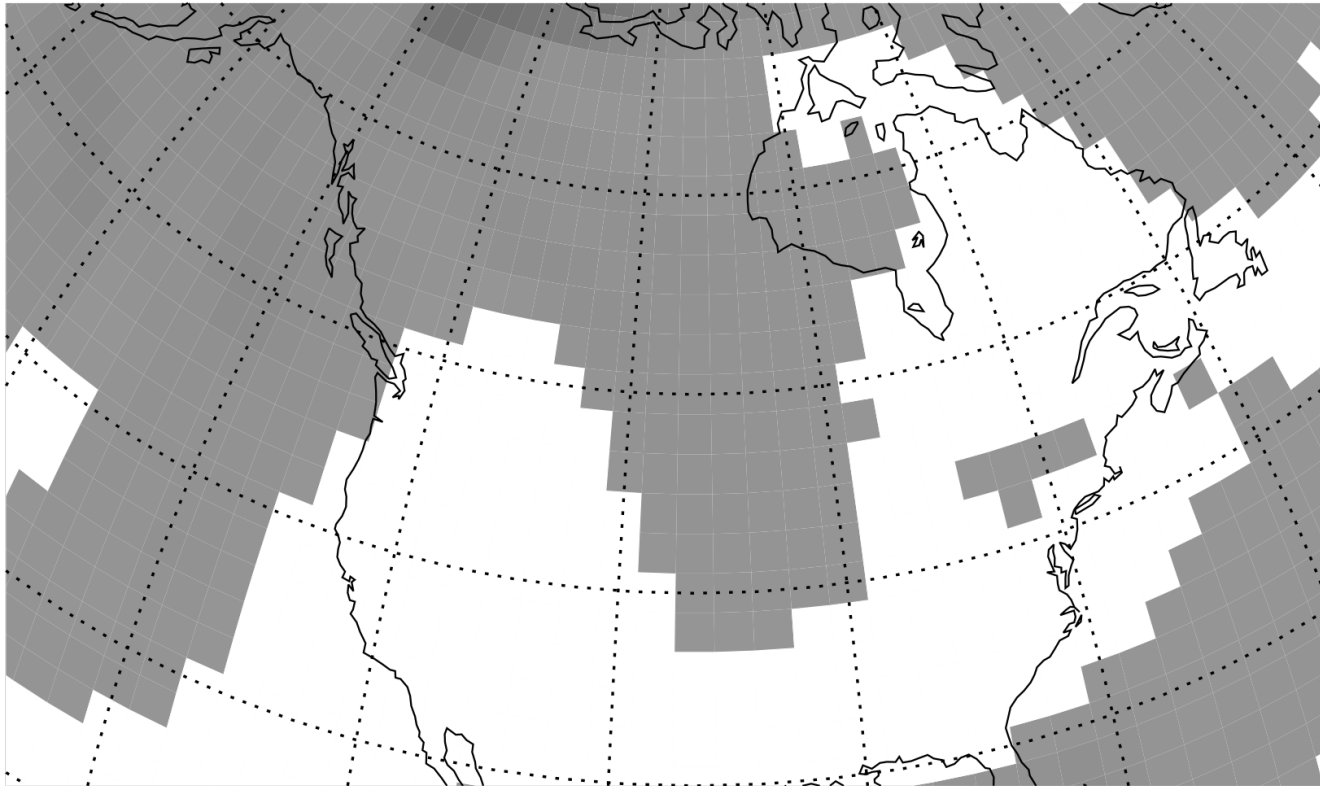
Siberian Smoke 20030503



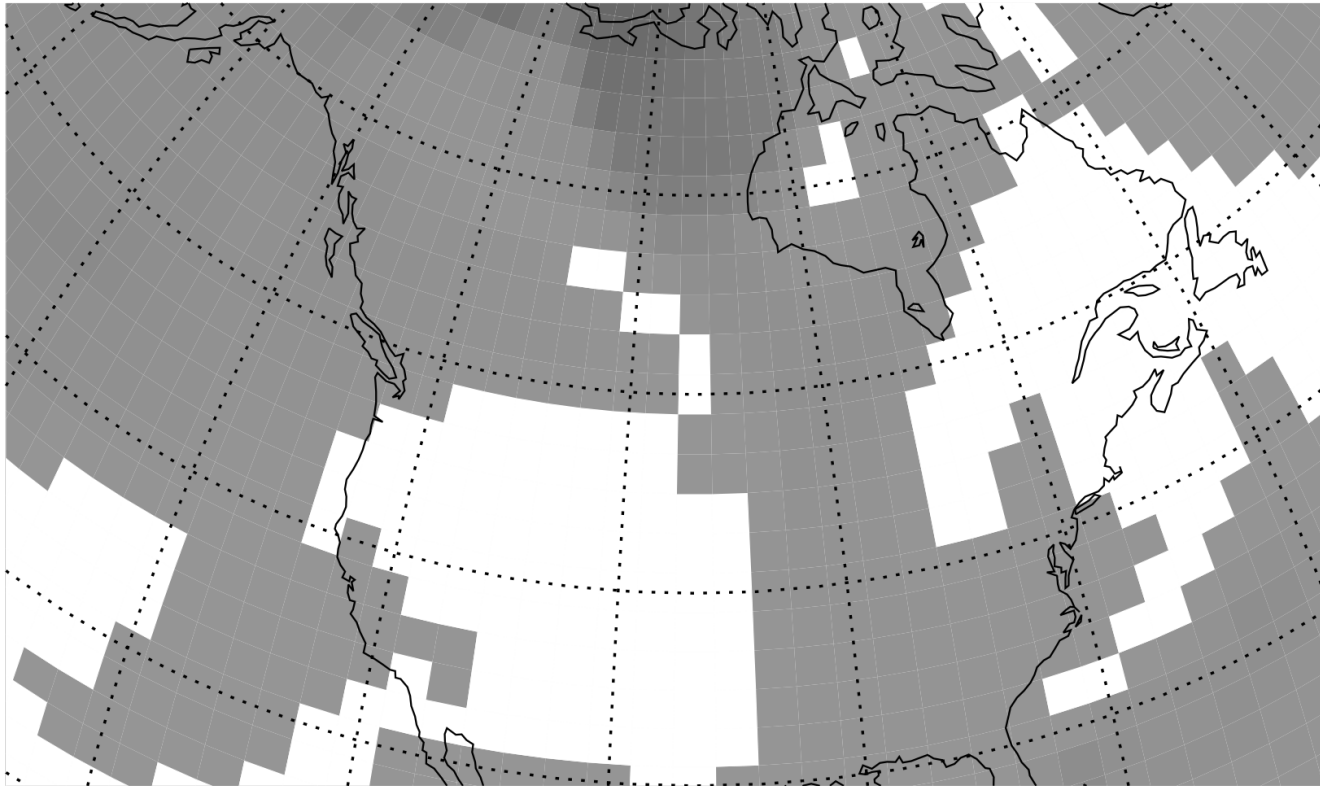
Siberian Smoke 20030527



Siberian Smoke 20030528



Siberian Smoke 20030529



Summary

- Four major aerosol types: dust, sea salt, sulfate and carbonaceous
- Aerosol cool the Earth by scattering, AKA the direct aerosol effect
- Aerosol warm the Earth by absorbing and forcing cloud evaporation, AKA the semi-direct aerosol effect
- Aerosol cool the Earth by making clouds more reflective, AKA the indirect aerosol effect

Mie Scattering

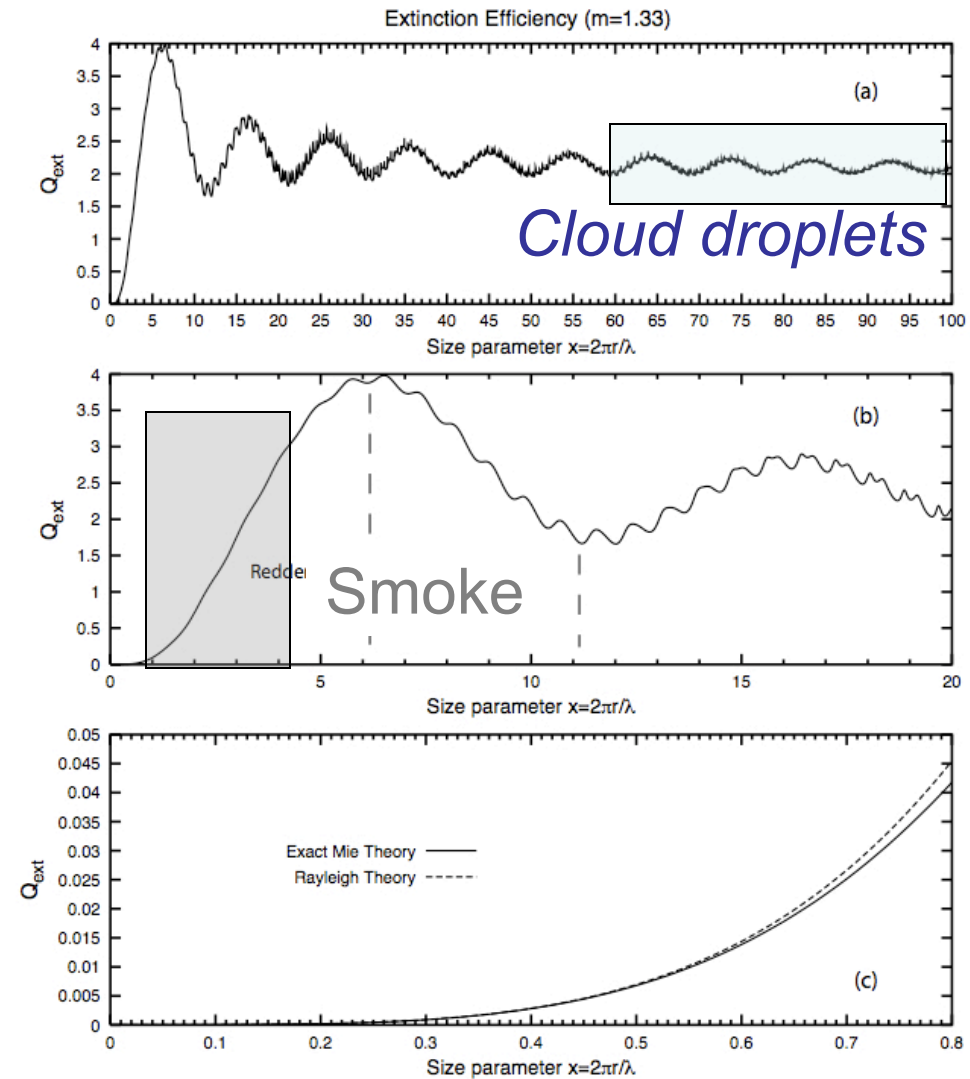


Fig. 12.4: The extinction efficiency Q_e as a function of size parameter x for a non-absorbing sphere with $m = 1.33$, for various ranges of x . (a) "Big picture" view, showing that $Q_e \rightarrow 2$ as $x \rightarrow \infty$. (b) Detail for $x < 20$, with examples of subranges for which extinction increases with x (reddening) or decreases with x (blueing). (c) Detail for $x < 0.8$, comparing the Rayleigh (small particle) approximation and exact Mie theory.

Absorption of Sunlight and Emission of Heat

