

Climate Change VS. Smog

The Story

- California Dairy Farmer, John Fiscalini, has installed a methane capturing system that generates enough energy to sustain his farm.
- Methane capturing dairy farms in San Joaquin Valley being shut down.
- The combustion engine used to generate the electricity contributes to smog.
- The Federal air quality regulations in California over their limits, making the cut back of smog more important than reduction of climate change.
- NO_x is associated with combustion engines like the methane generated engines on the dairy farm.
- NO_x regulations are 11ppm
- NO_x makes PM and ozone
- Animal Husbandry produces CH₄ and Ammonia a precursor to particulate matter
- According to a news source Methane generators potentially:
 - Reduce H₂S, VOC, and PM (How?)
 - Reduces Odor
- Point source Vs. Nonpoint source issue
- The installation of the generators on Fiscalini's farm costs 4 million dollars.
- The additional pollution controls on Fiscalini's farm cost 200,000 dollars.
- 2009 six generators shut down
- Produce less NO_x than coal power plants, but no coal power plants are in the San Valley Air Pollution Control District
- San Joaquin Valley rules say need Best Available Control Technology for the area
- San Joaquin Valley Air Pollution Control district is targeting reduction in VOC and NO_x

Textbook Chemistry

- Troposphere:
 - Oxidation of CH₄
 - O[•]D + H₂O → OH + OH
 - OH + CH₄ → CH₃ + H₂O
 - CH₃ + O₂ + M → CH₃O₂ + M
 - CH₃O₂ + HO₂ → CH₄O₂ + O₂ (dissolves in rain)
 - CH₃O₂ + NO → CH₃O + NO₂ (preserves radical while also a source for NO₂)
 - CH₃O + O₂ → CH₂O + HO₂
 - CH₂O + hv → H + HCO
 - HCO + O₂ → CO + HO₂
 - NO_x production of Ozone
 - NO₂ + hv → O + NO
 - O + O₂ + M → O₃ + M

- $\text{OH} + \text{CO} \rightarrow \text{H} + \text{CO}_2$
 - $\text{H} + \text{O}_2 + \text{M} \rightarrow \text{HO}_2 + \text{M}$
 - $\text{HO}_2 + \text{NO} \rightarrow \text{OH} + \text{NO}_2$ (production of ozone and source for NO_2)
 - $\text{NO}_2 + \text{O}_3 \rightarrow \text{NO}_3 + \text{O}_2$ (sink for ozone, but NO_3 photolysis during day)
 - Peroxyacetyl nitrate (PAN)
 - $\text{CH}_3\text{COO}_2(\text{water soluble}) + \text{NO}_2 \rightarrow \text{CH}_3\text{COO}_2\text{NO}_2$ (PAN found in photochemical smog)
 - PAN can cause asthma, chest pain, headaches, coughing, asthma attacks
 - Heterogeneous Nox chemistry
 - $\text{NH}_3(\text{g}) + \text{HNO}_3(\text{g}) \rightarrow \text{NH}_4\text{NO}_3(\text{s})$
 - $\text{NaCl} + \text{HNO}_3 \rightarrow \text{HCl}(\text{g}) + \text{NaNO}_3$ (condensation)
- CH_4 a Green House Gas
 - Long wavelength energy absorbed by CH_4 increasing global temperature

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